



ASMS Fall Workshop 2004

Polymer Structure

Baltimore Dec. 9-10



Max Planck Institute for Polymer Research

Polymer Structure

elucidated by

Mass Spectrometry

Hans Joachim Räder



ASMS Fall Workshop 2004

Polymer Structure

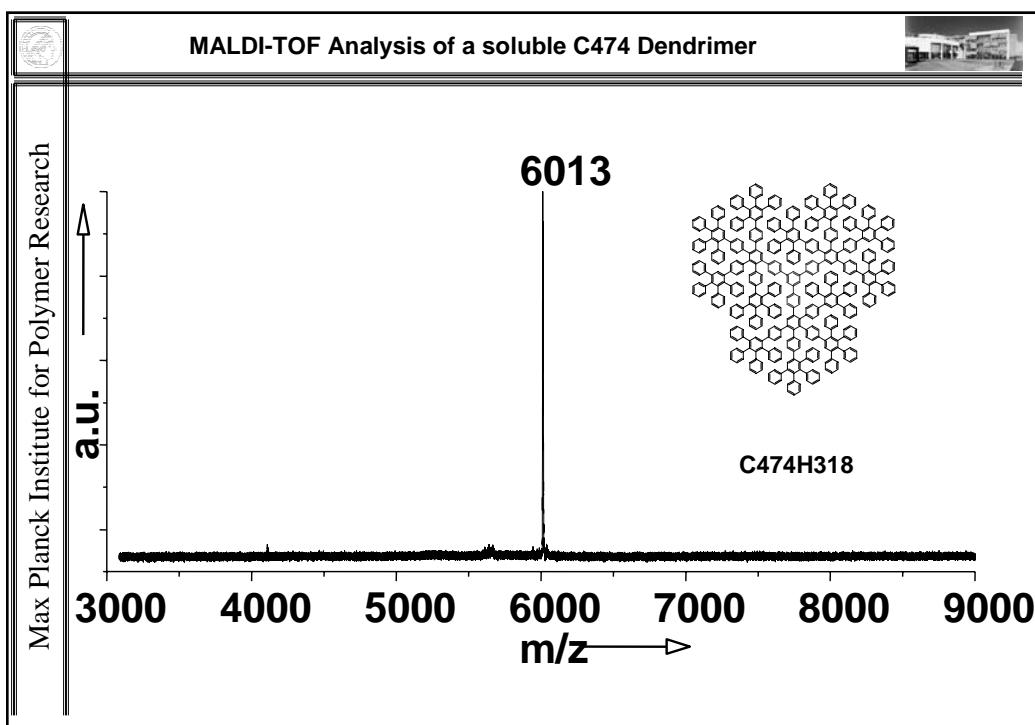
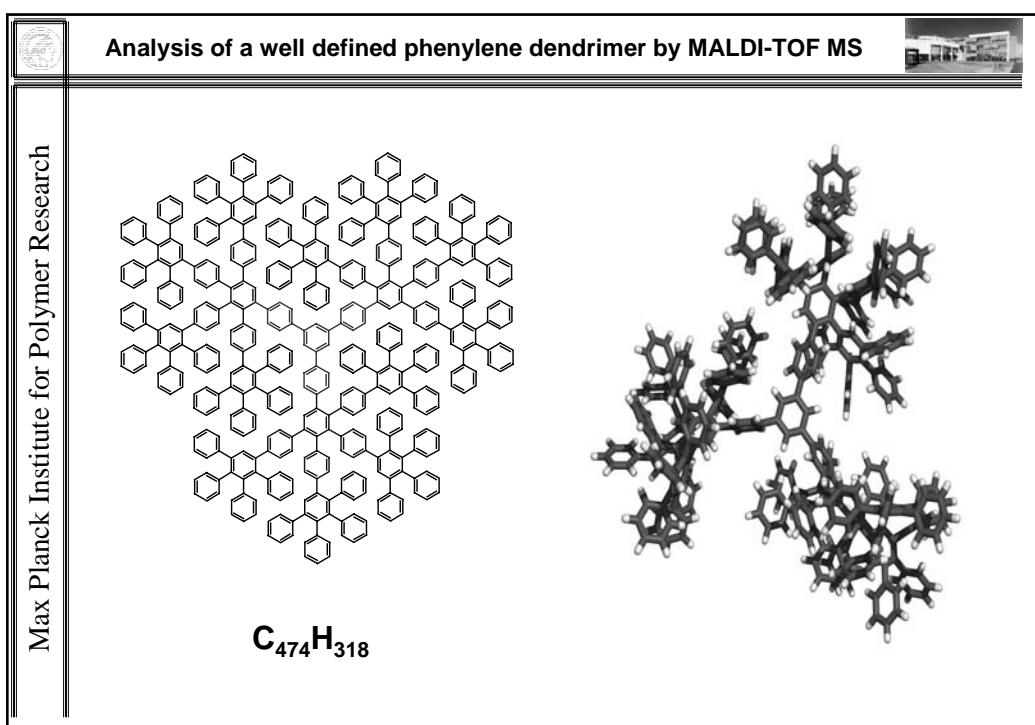
Baltimore Dec. 9-10

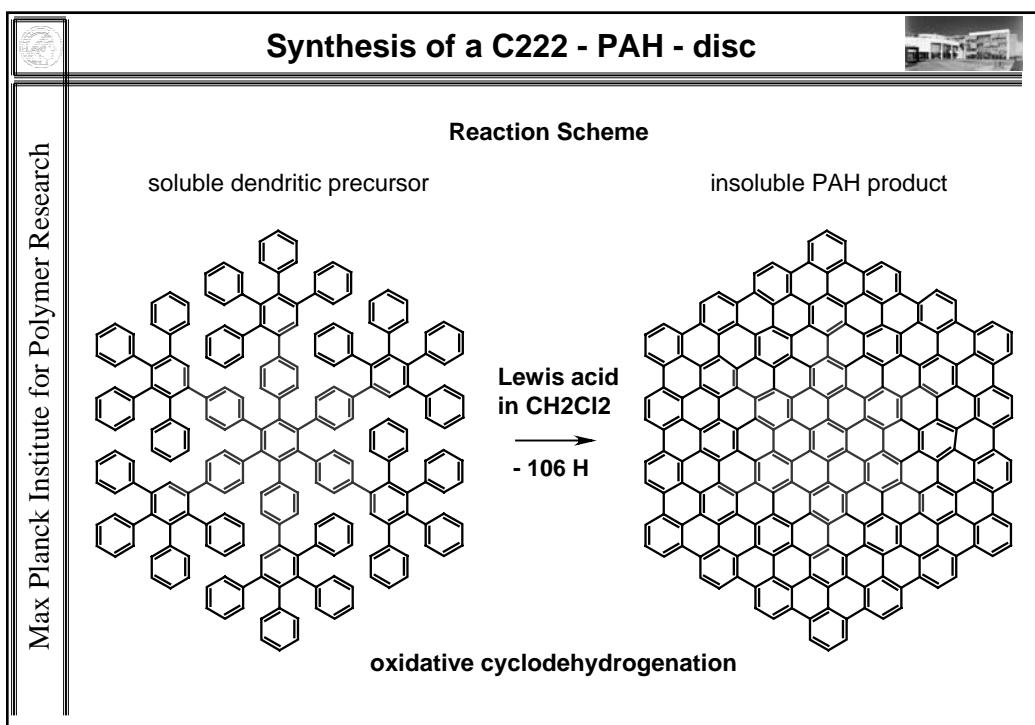
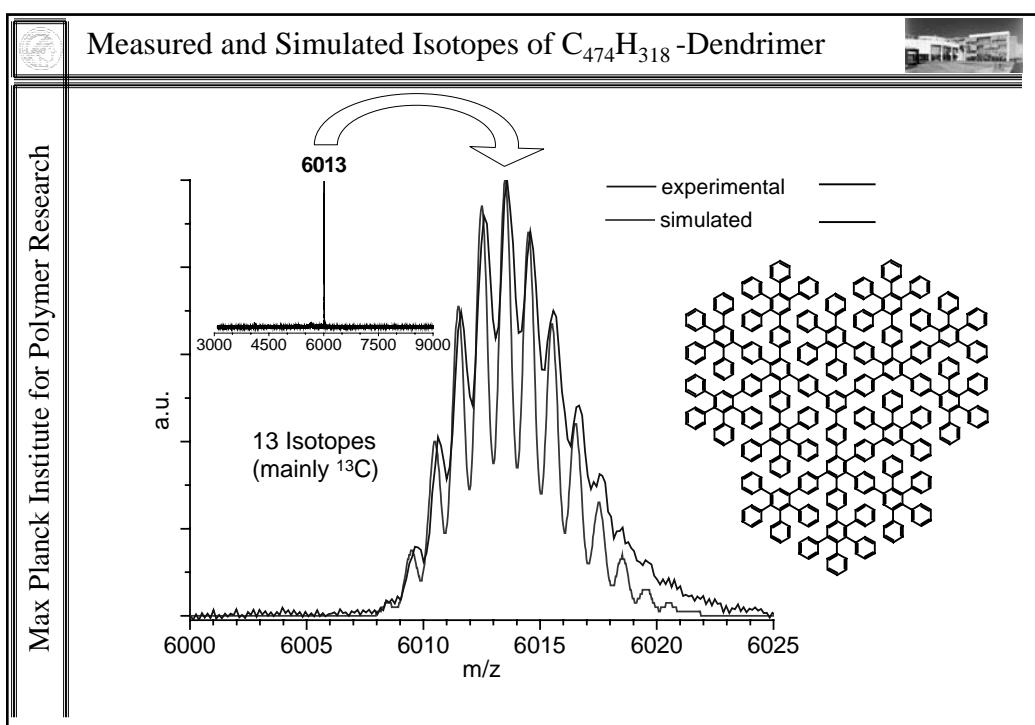


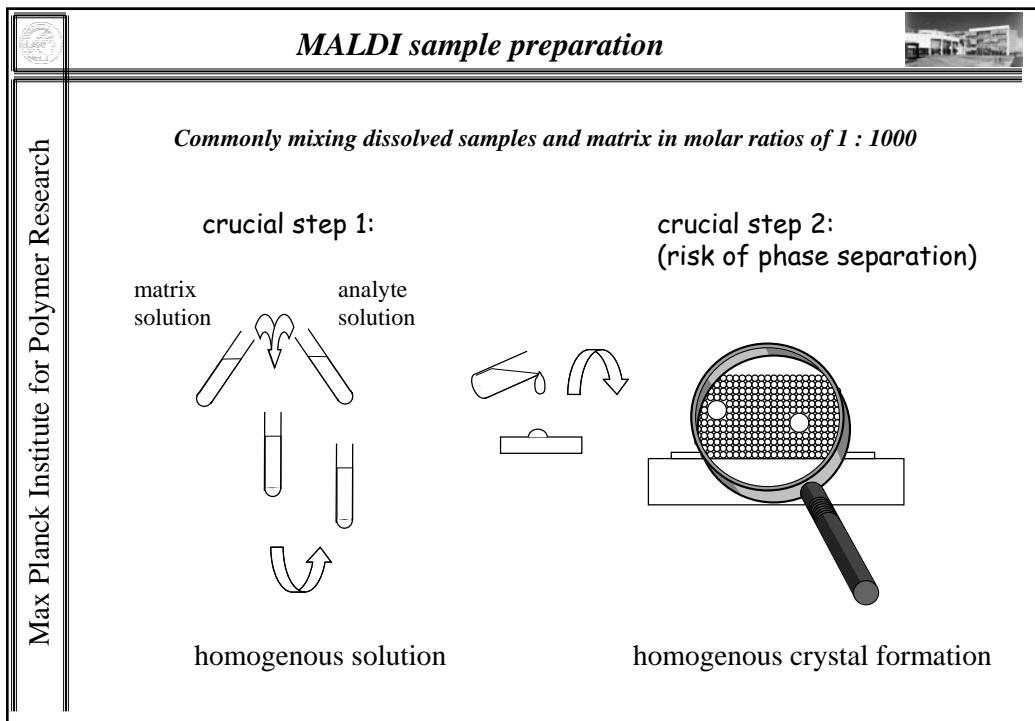
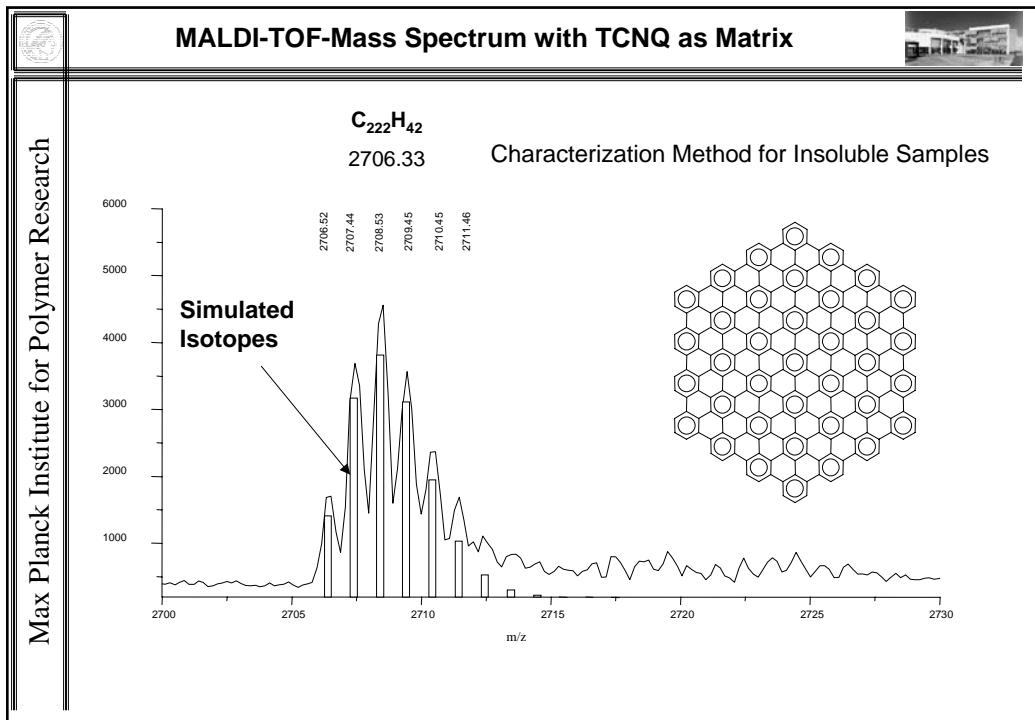
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Outline

- Structure determination of monodisperse macromolecules
 - Phenylene dendrimers
 - Giant polycyclic aromatic hydrocarbons (PAH's)
- Structure determination of polydisperse macromolecules
 - End group determination of homo-polymers
 - in simple cases (Poly(styrene))
 - in difficult cases (Poly(carbonate) and Poly(fluorene))
- Copolymers
 - PPE-b-PEO Diblock-copolymer
 - Random copolymers of amino acids









Solvent-free sample preparation

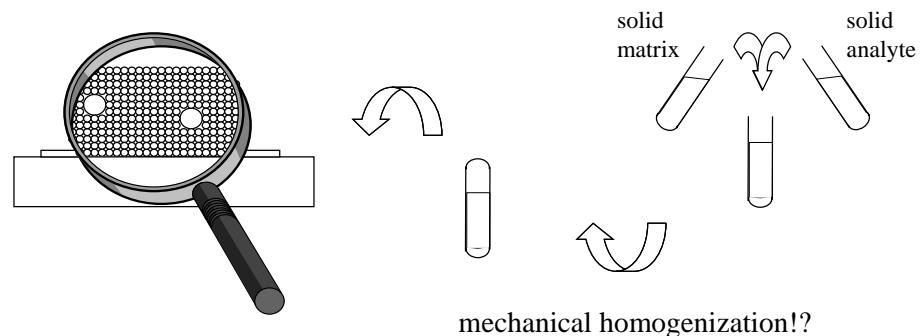


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Mechanical mixing of solid samples in e.g. a ball mill

crucial step 2 is overcome!
No risk of phase separation

crucial step 1:

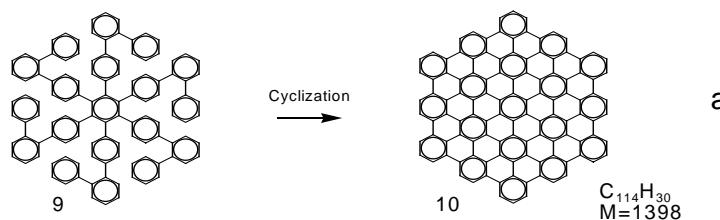


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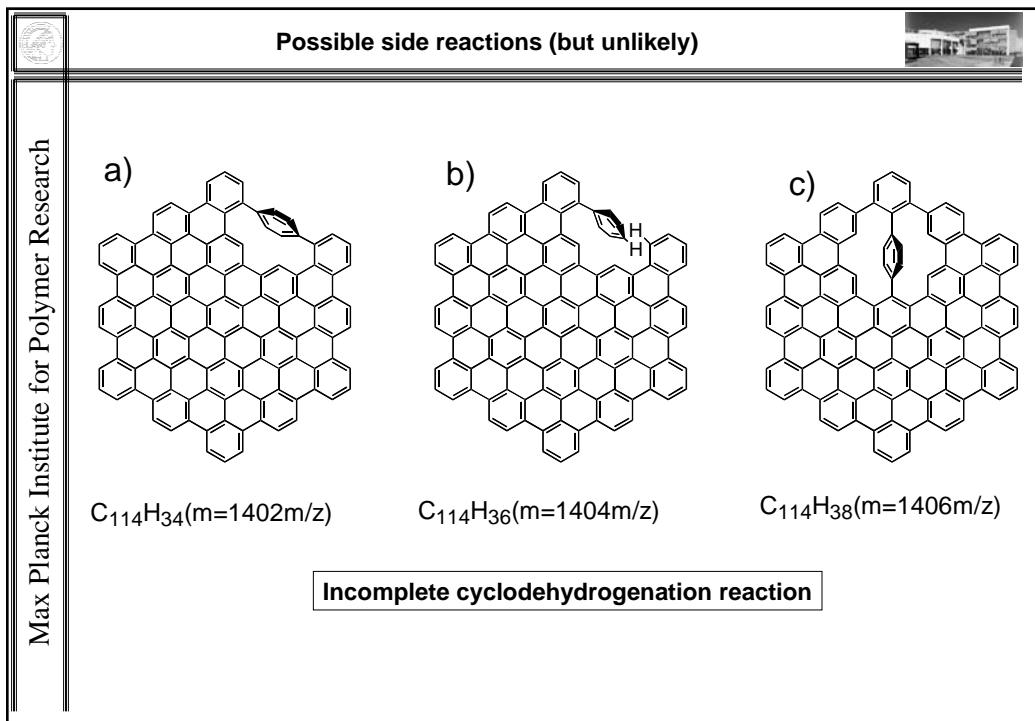
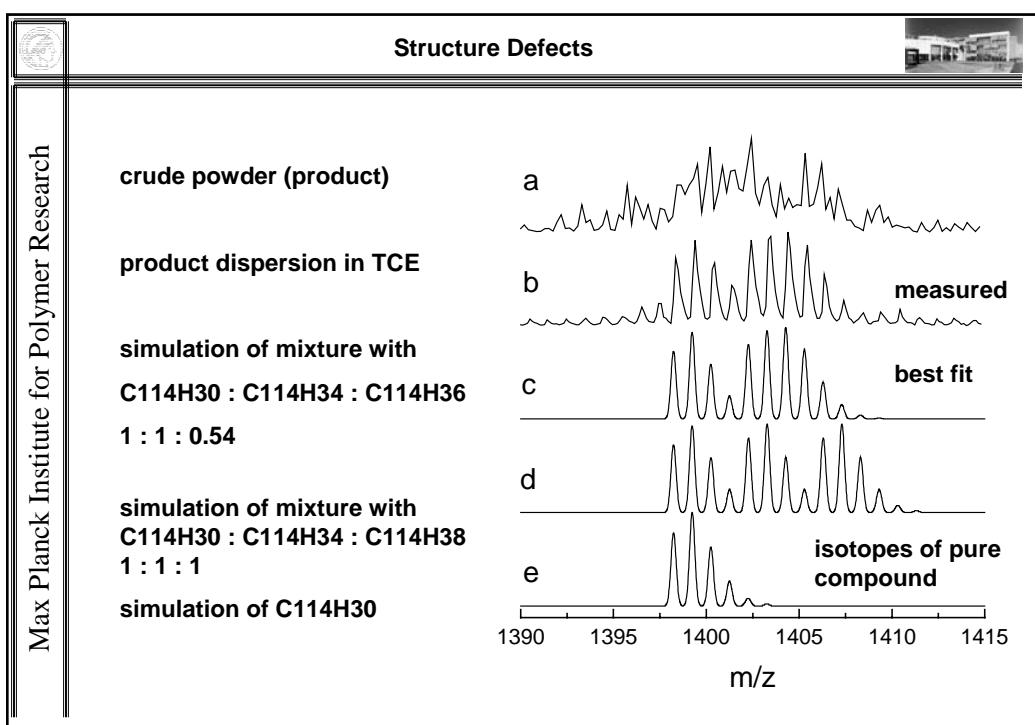
Structure Defects

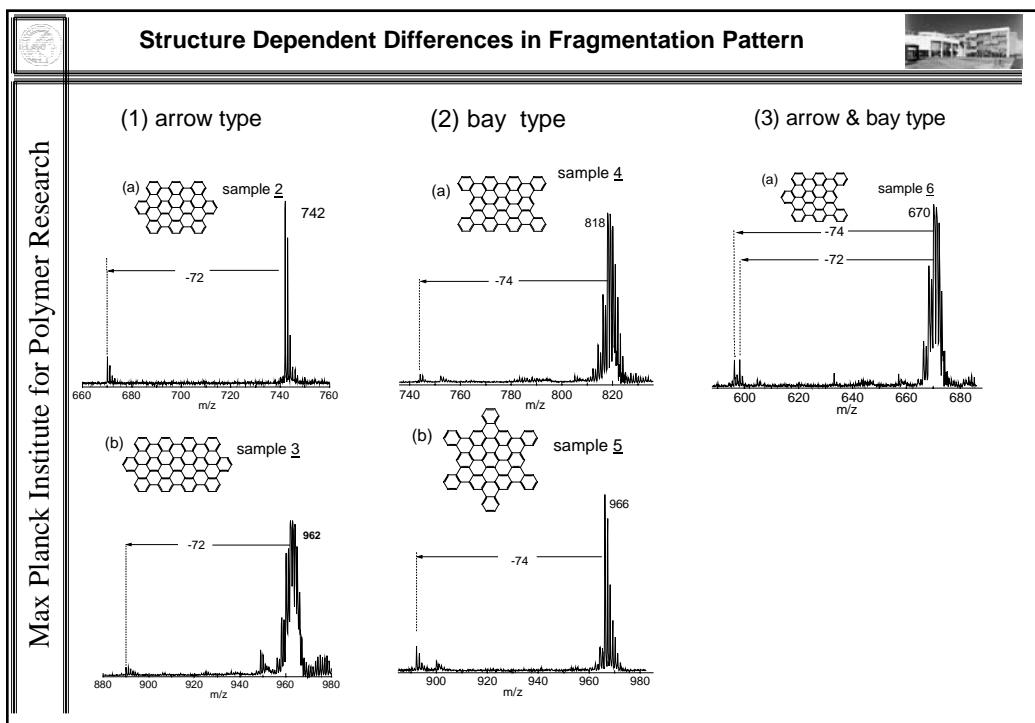
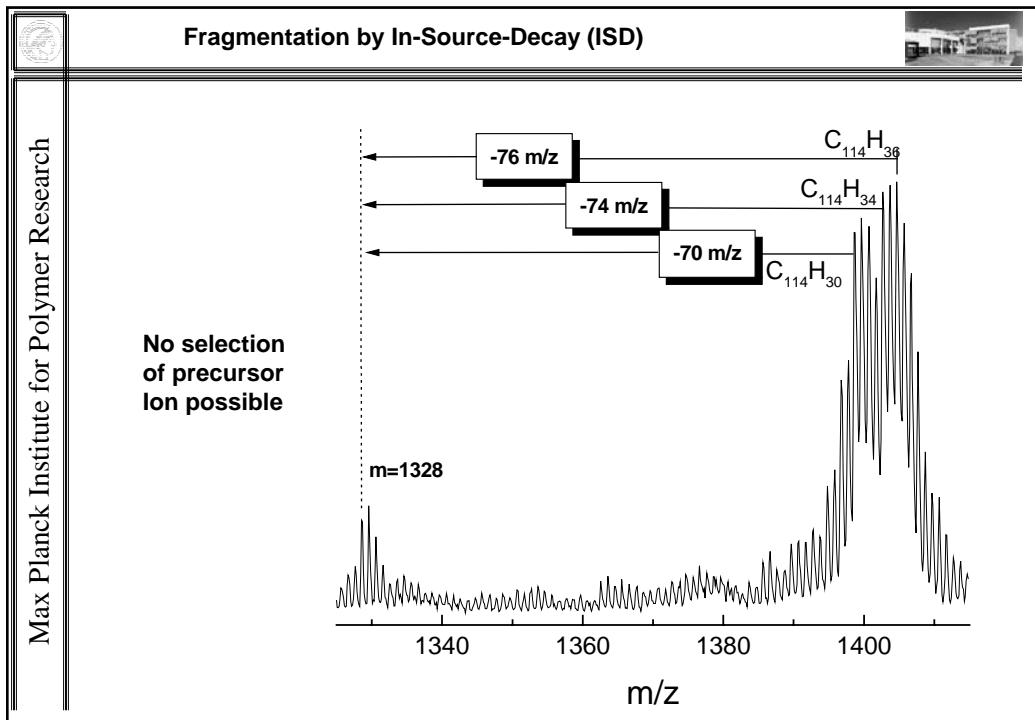


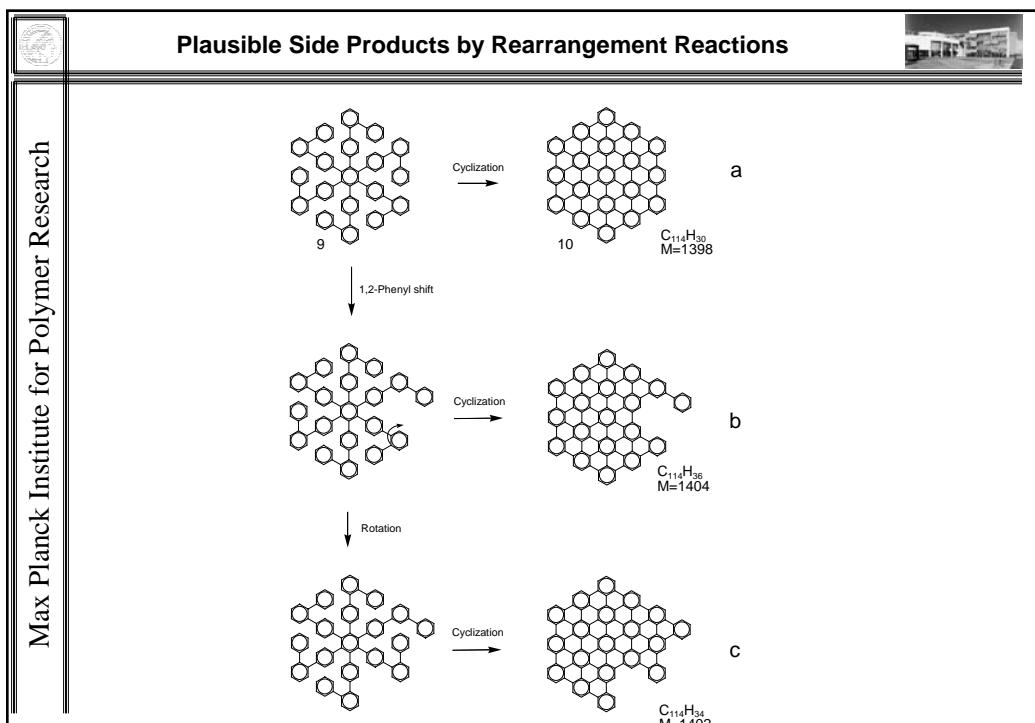
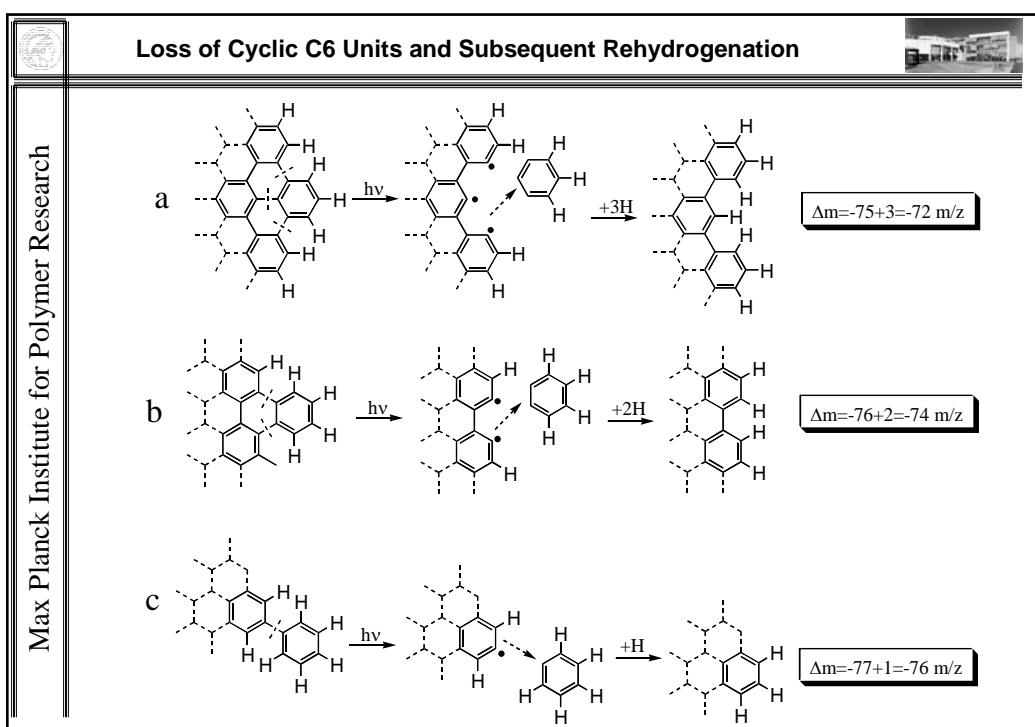
Product mixture with different numbers of hydrogens

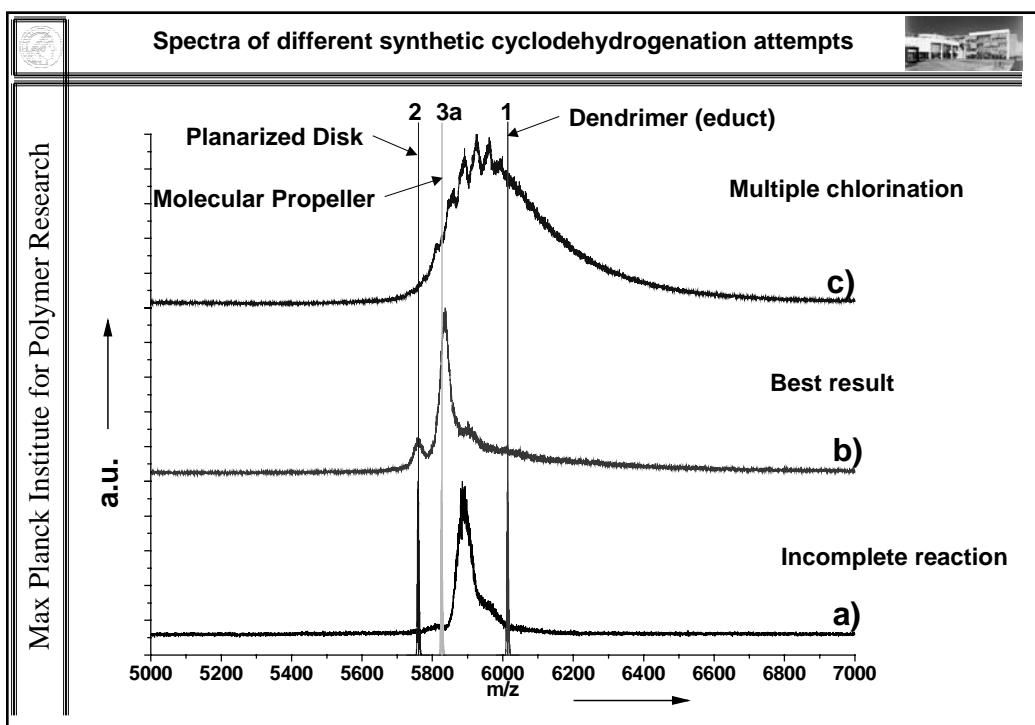
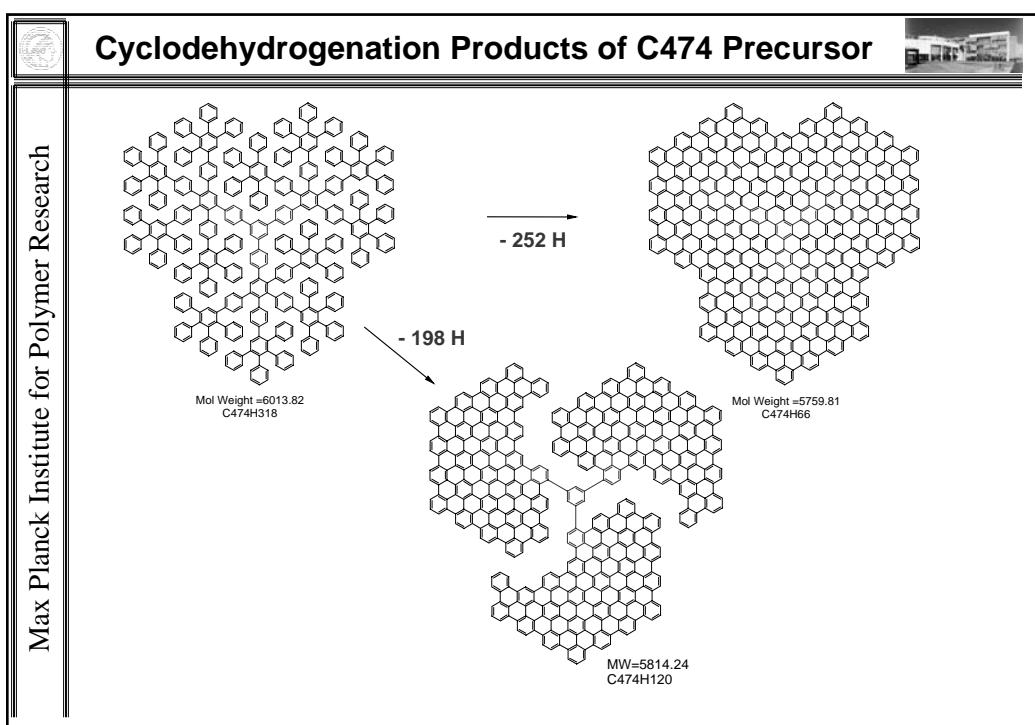


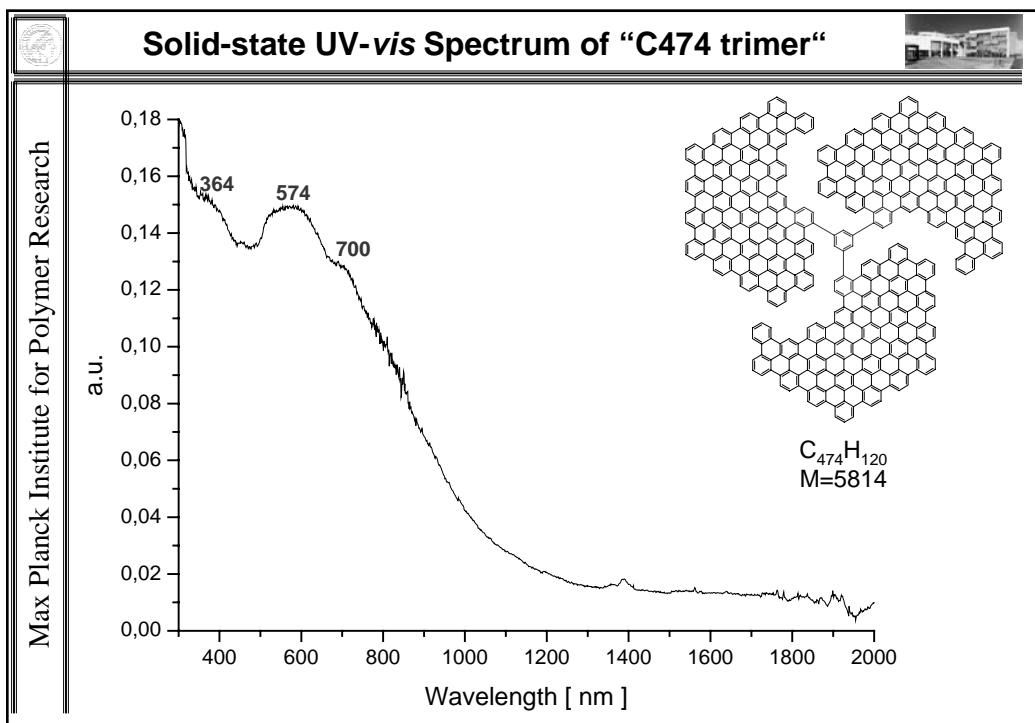
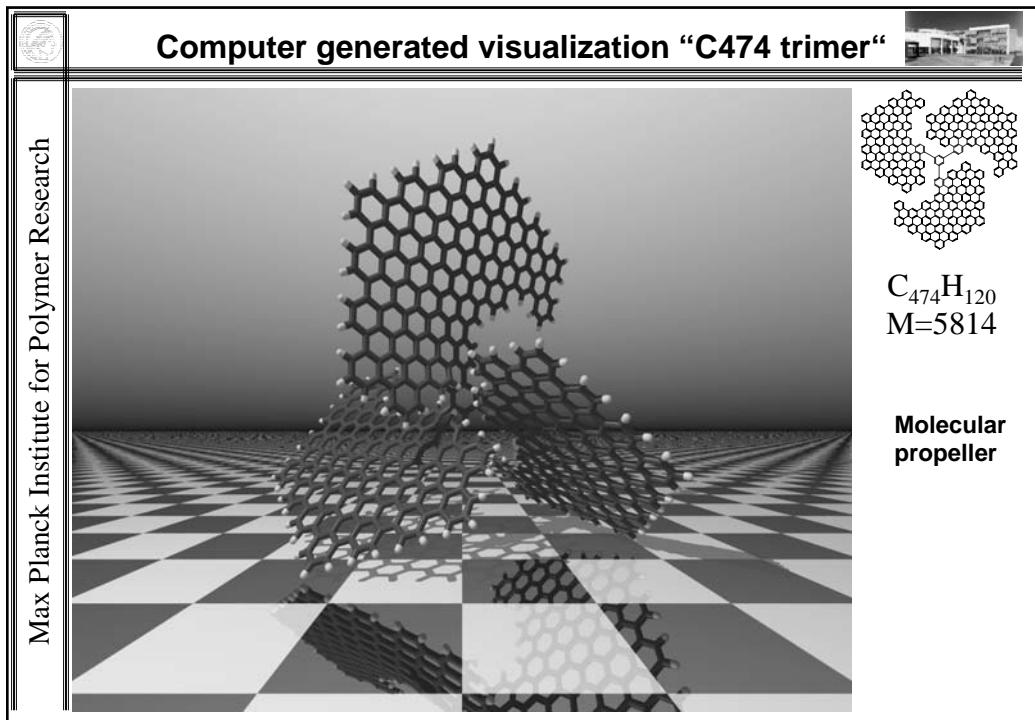
Incomplete Cyclodehydrogenation?

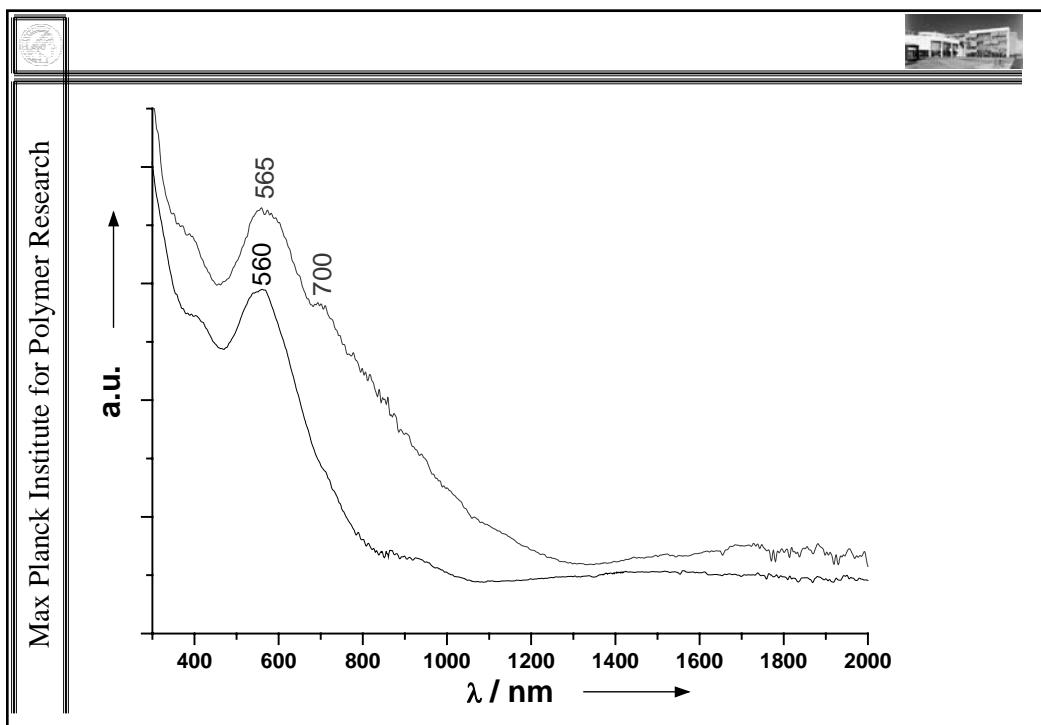
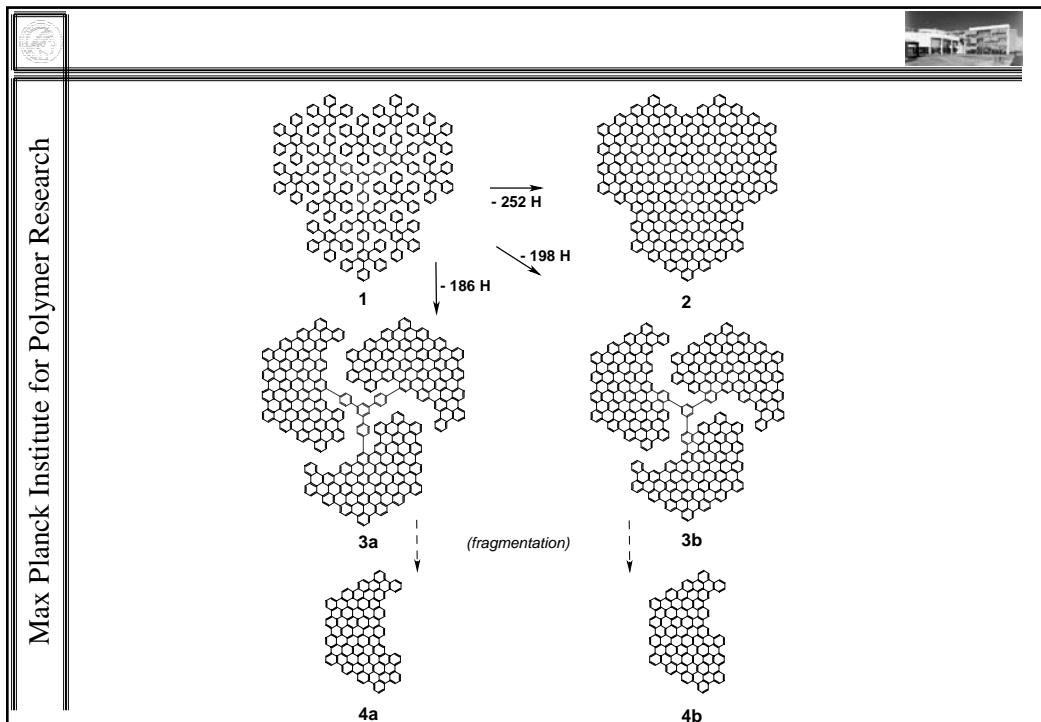








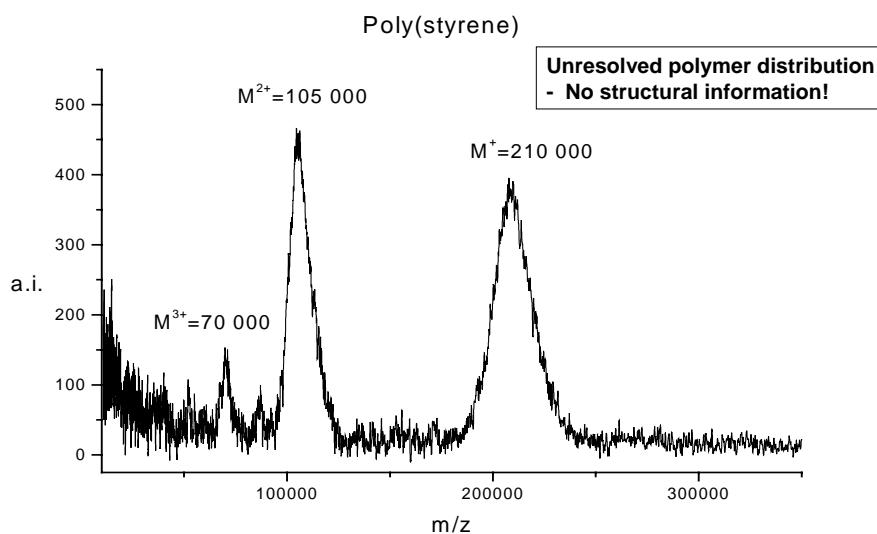


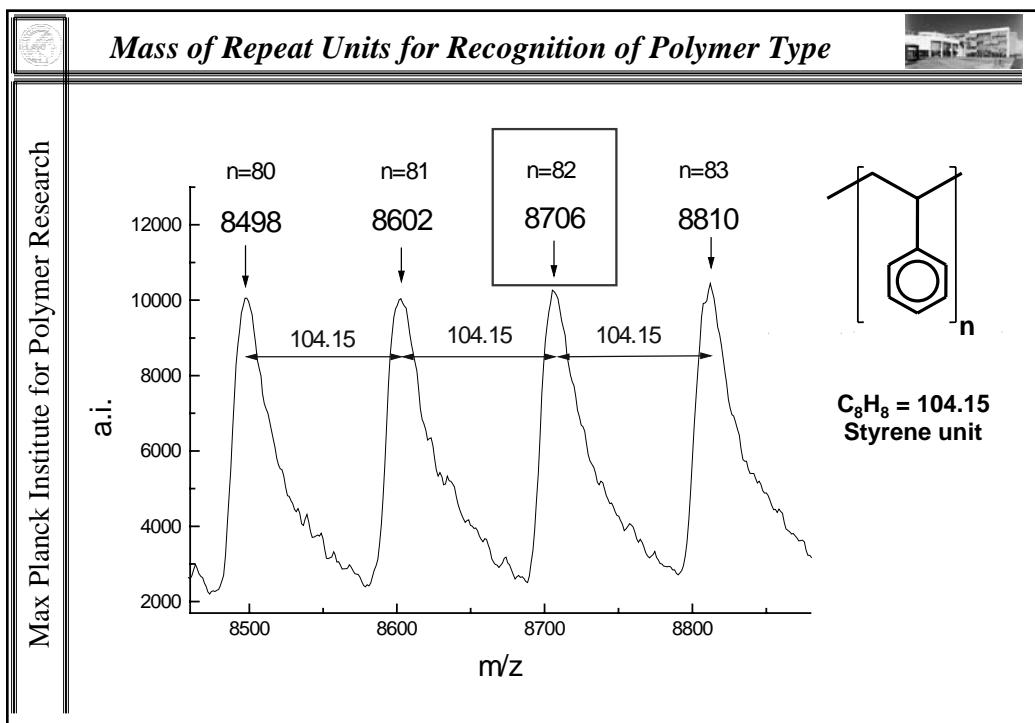
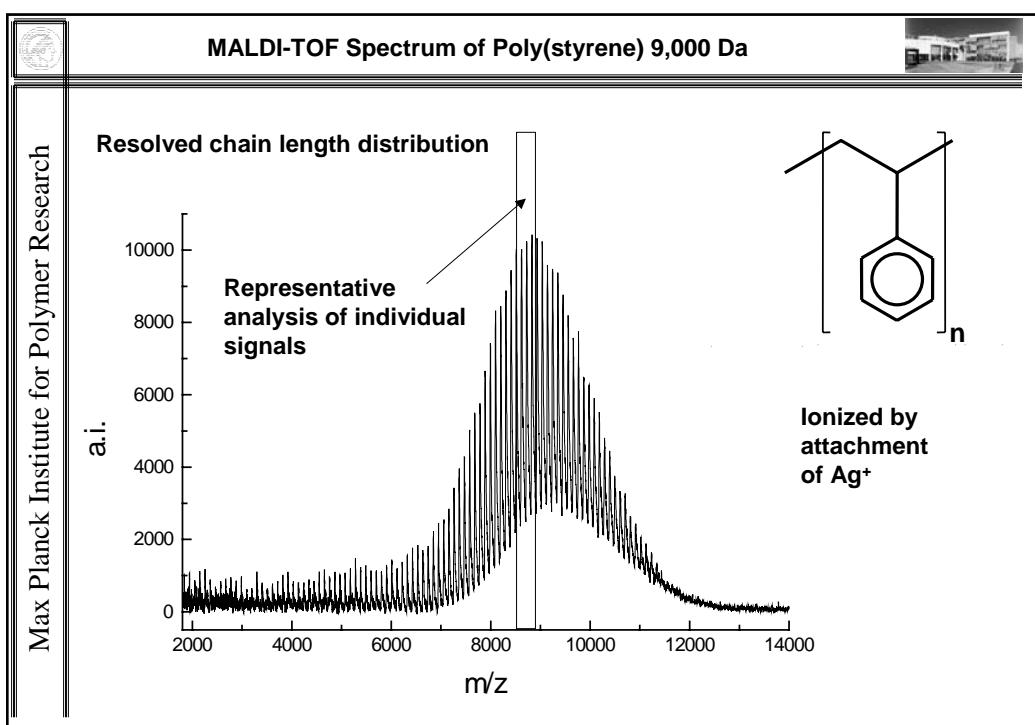




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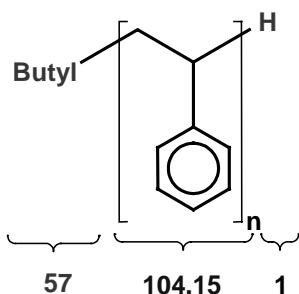




Calculation of End Group Molecular Weight



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Mass contributions:

$$8706 = 82(n) * 104.15 \text{ (Styrene)} \\ + 57 \text{ (Butyl)} \\ + 1 \text{ (Hydrogen)} \\ + 108 \text{ (Ag}^+)$$

End group calculation:

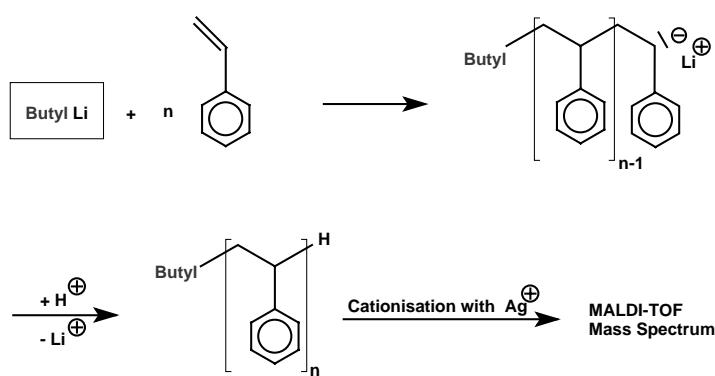
$$8706 - 108 \text{ (Ag}^+) = 8598 \text{ (corrected polymer mass)} \\ 8598 : 104.15 \text{ (styrene unit)} = 82.554 \text{ (repeat units)} \\ 82.554 \longrightarrow 82 \text{ styrene repeat units (n)} \\ \text{and decimal place bears end group mass:} \\ 0.554 * 104.15 = 57.7 \text{ (mass of both end groups)}$$

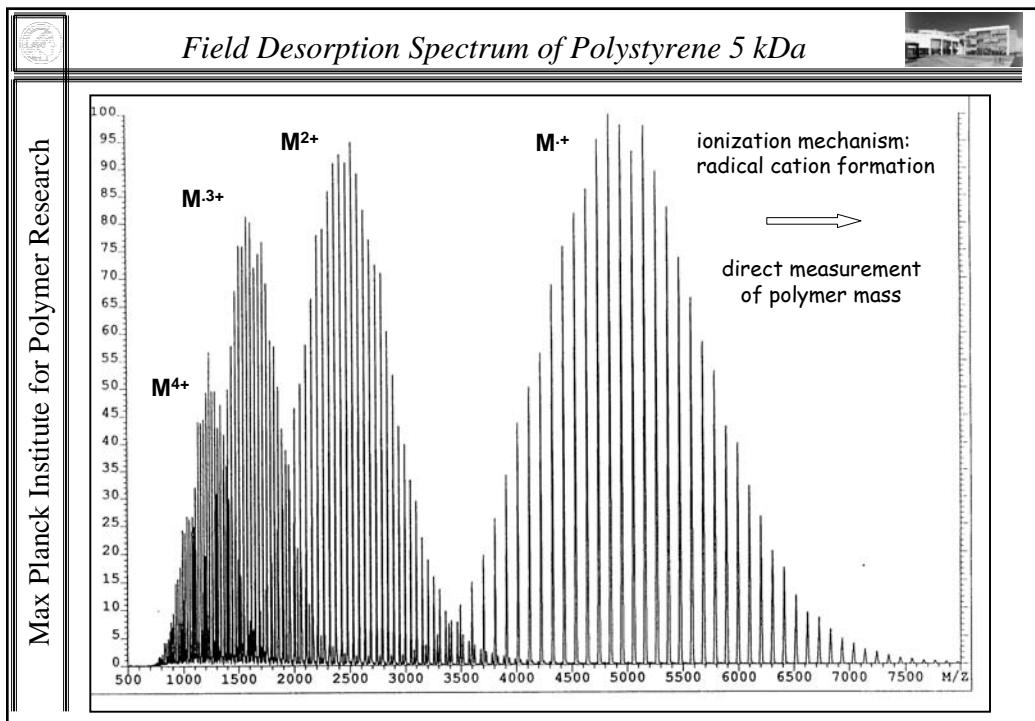
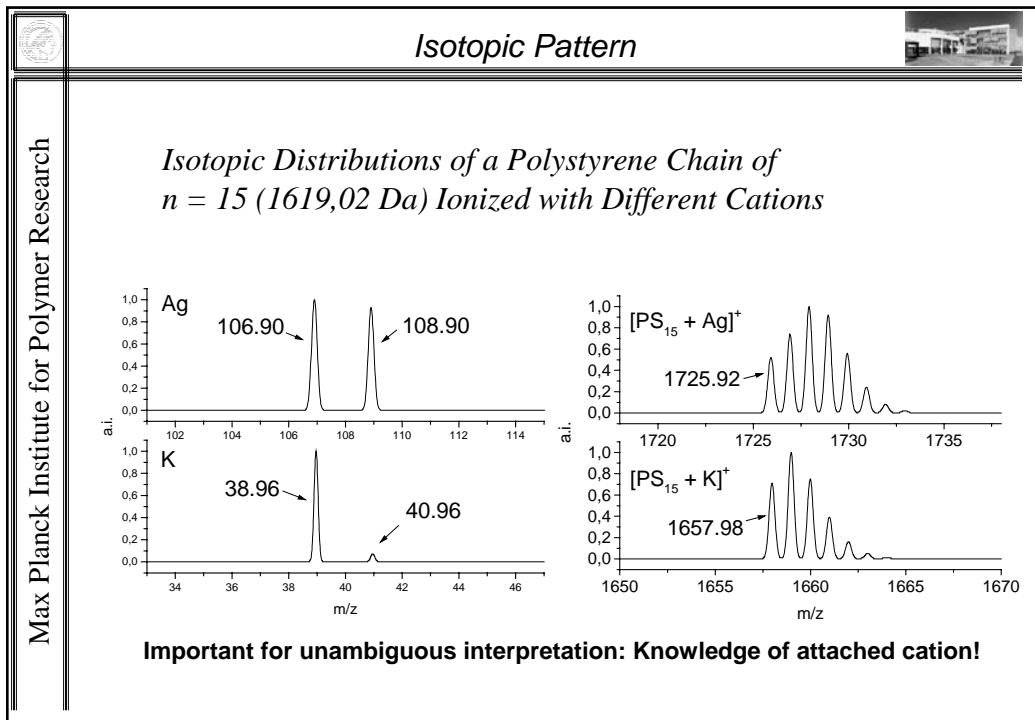


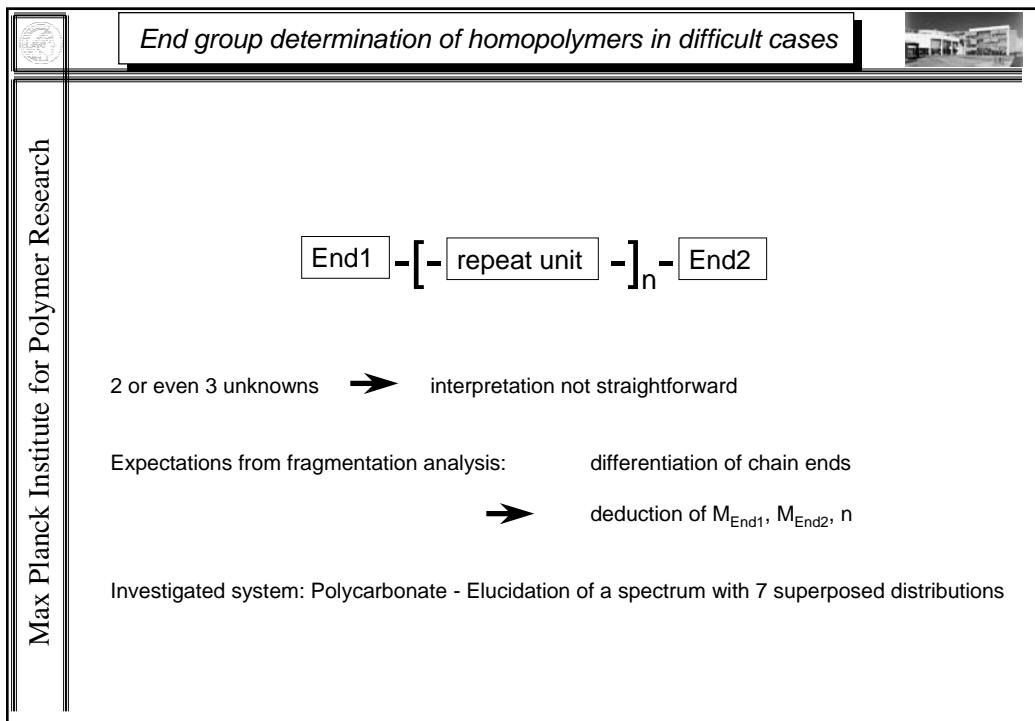
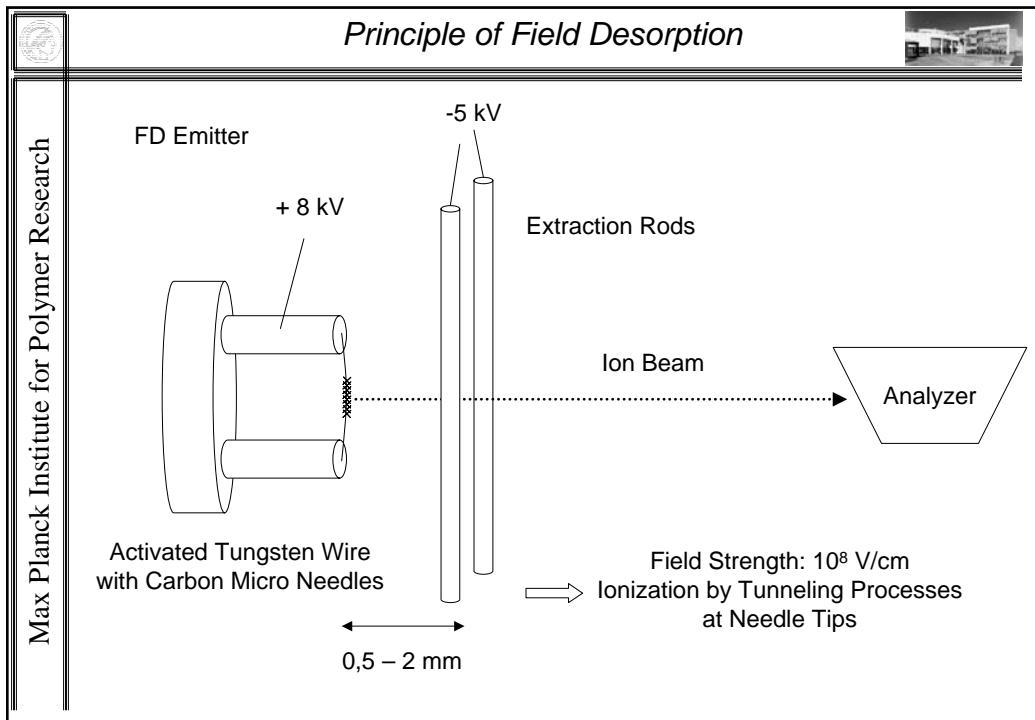
Sample History: Way of Synthesis

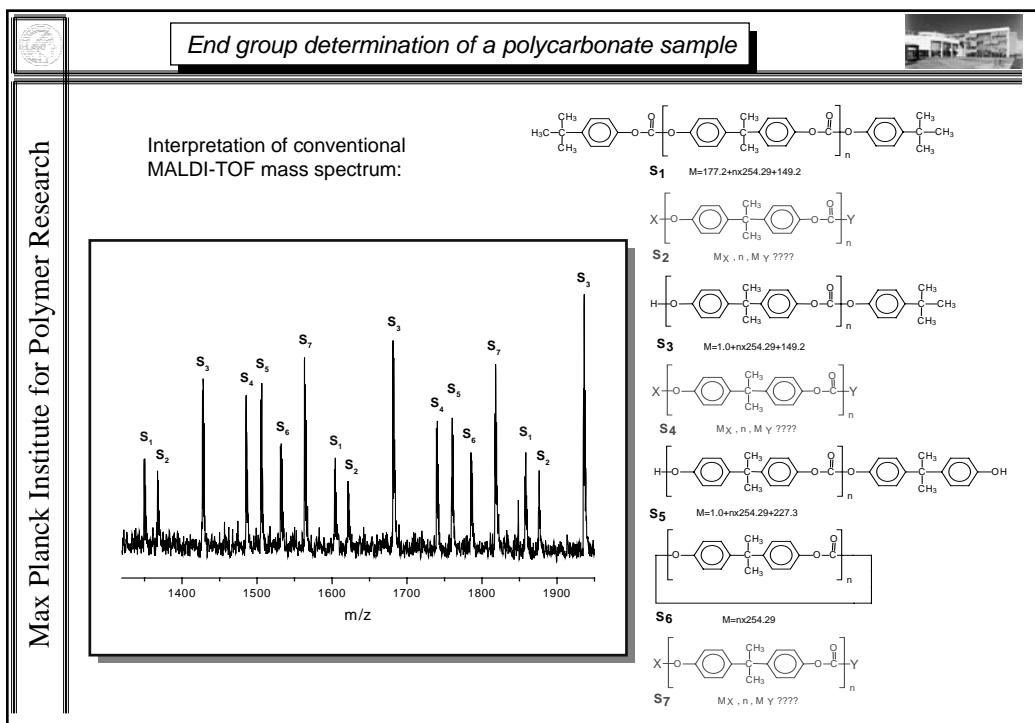
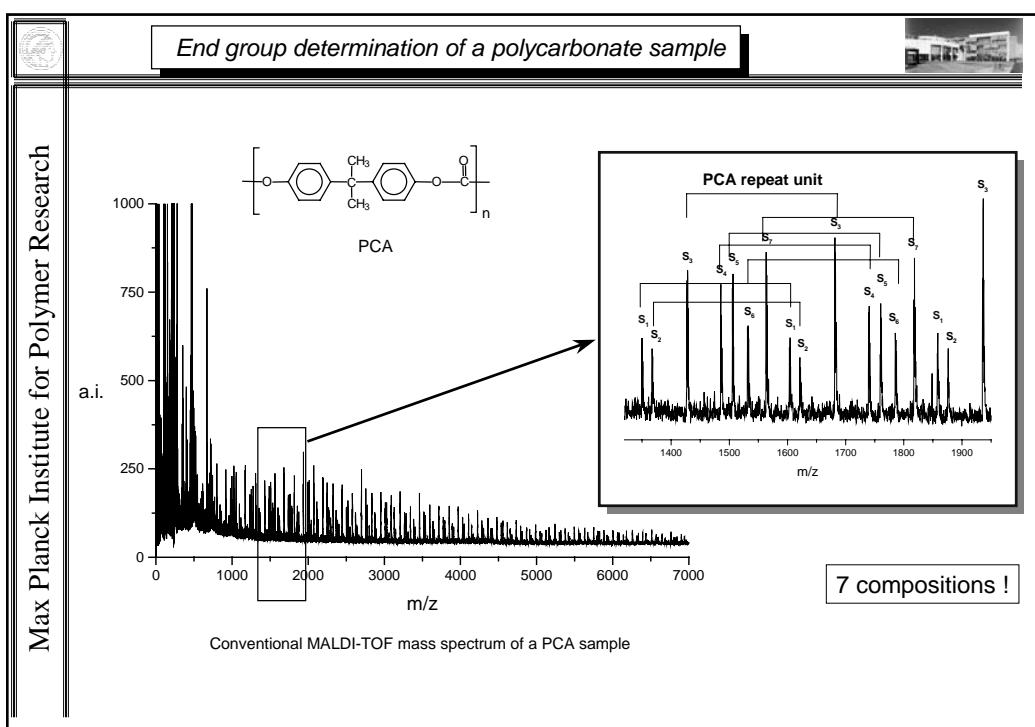
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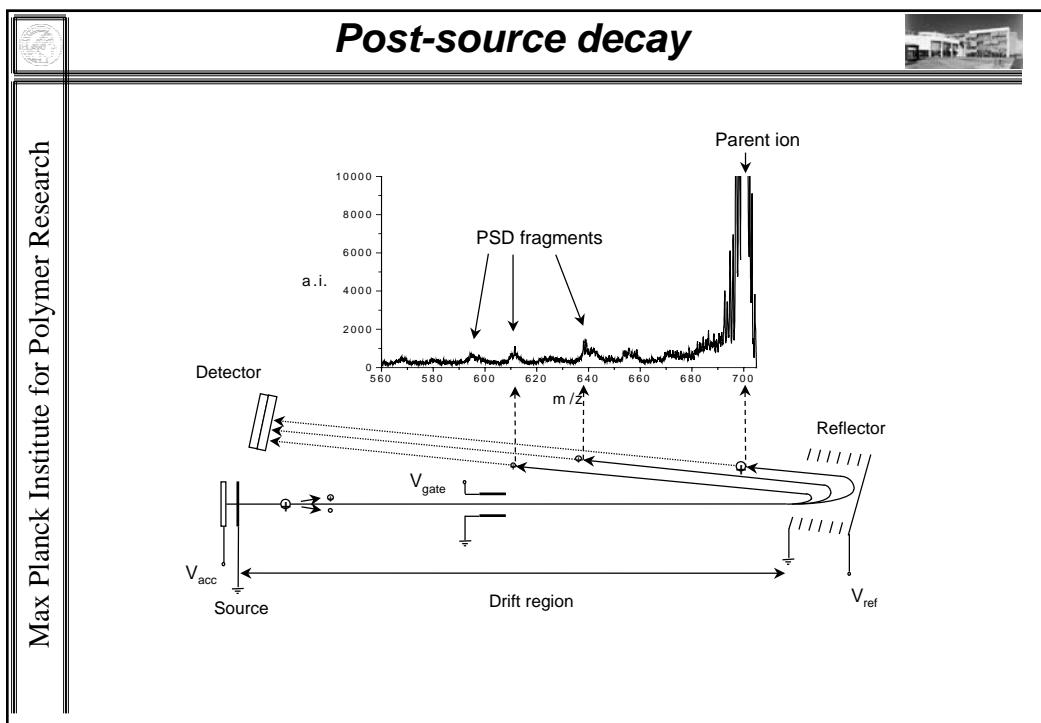
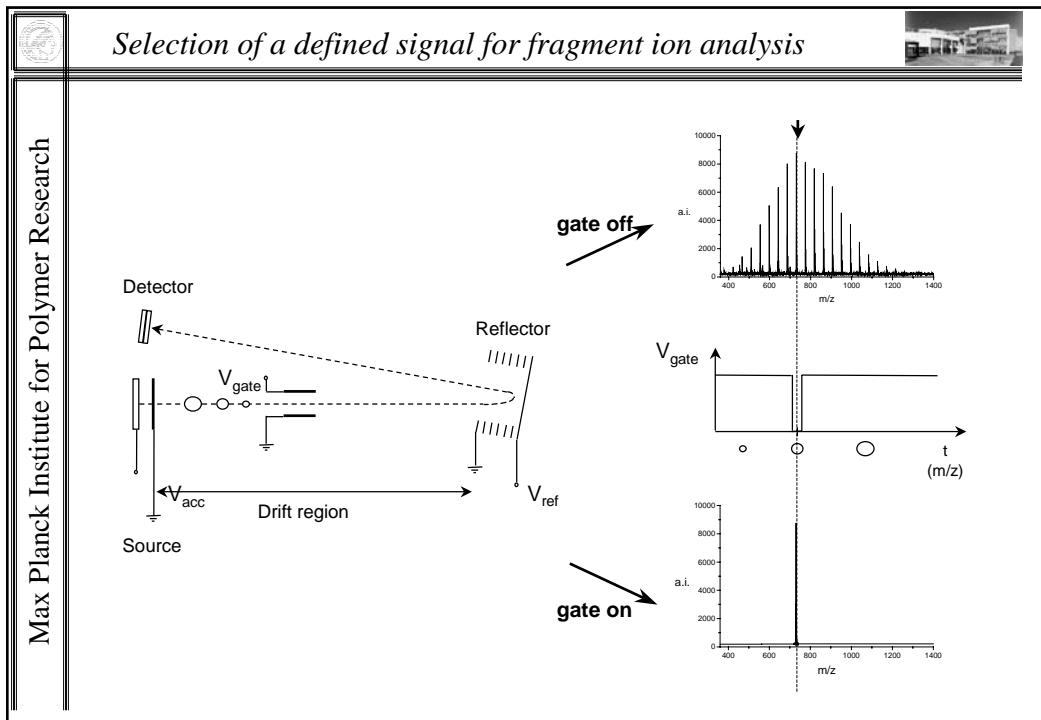
Anionic Polymerization of Styrene











Polycarbonate: fragmentation mechanism?

Main fragmentation of PCA:

The top part shows the chemical structure of the polycarbonate repeating unit: $\left[\text{CH}_3\text{C}(\text{CH}_3)\text{O}(\text{C}_6\text{H}_4)\text{OC}(=\text{O})\text{O}(\text{C}_6\text{H}_4)\text{C}(\text{CH}_3)\text{CH}_3 \right]_n$. Below it, two fragmentation pathways are shown starting from the repeating unit ion $[\text{Li}^+]$:

- $- \text{CH}_3$ loss leads to the fragment $[\text{Li}^+]$ with the structure $\left[\cdots \text{C}_6\text{H}_4\text{COO}(\text{C}_6\text{H}_4)\text{C}(\text{CH}_3)\text{CH}_3 \right]$.
- $- \text{COO}(\text{C}_6\text{H}_4)$ loss leads to the fragment $[\text{Li}^+]$ with the structure $\left[\cdots \text{C}_6\text{H}_4\text{C}(\text{CH}_3)\text{CH}_3 \right]$.

analogue to mechanism described for EI, SSIMS (Literature)

Fragment mass spectrum:

Fragment mass spectrum

$\Delta M = M_{\text{CH}_3} + M_{\text{End}}$

$= 15 + M_{\text{End}}$

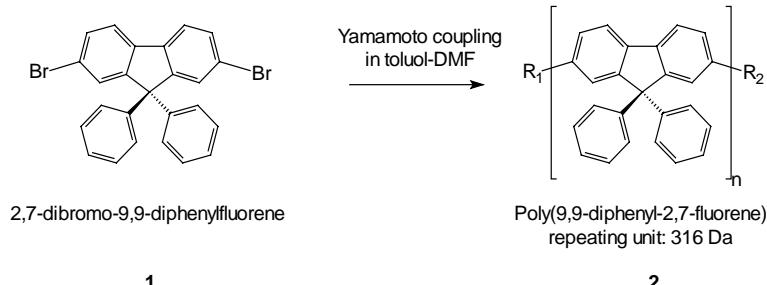
\rightarrow M_{End} deduced !



Synthesis of Polyfluorene (PF)



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2,7-dibromo-9,9-diphenylfluorene

1

Yamamoto coupling
in toluol-DMF

Poly(9,9-diphenyl-2,7-fluorene)
repeating unit: 316 Da

2

Dr. A. C.
Grimsdale

- 2a:** R₁=Br; R₂=Br
- 2b:** R₁=Br; R₂=H or R₁=H; R₂=Br
- 2c:** R₁=H; R₂=H
- 2d:** R₁=R₂=no atom (cyclic structure)

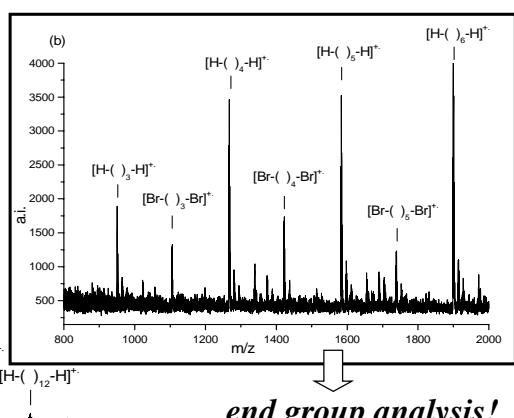
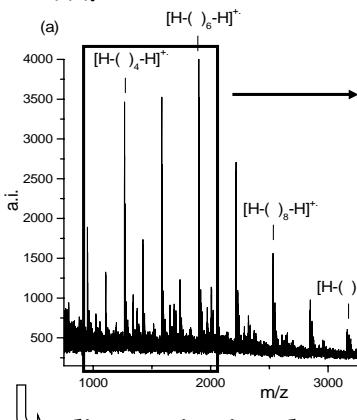


Solvent-Based MALDI MS of the Soluble Fraction (< 10%) of PF



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- dithranol
- THF



end group analysis!

oligomerization degree ~ n = 15

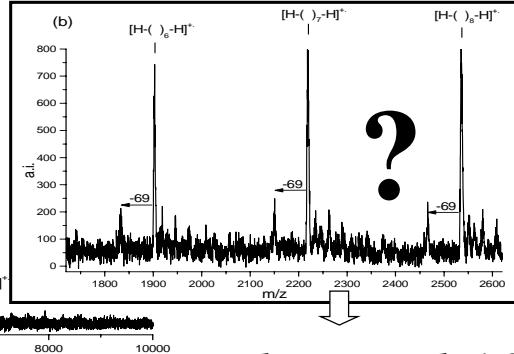
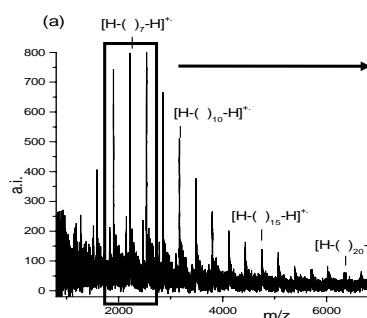


Solvent-Free MALDI MS of Insoluble Polymer Fraction (> 90%) of PF



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dithranol matrix



end group analysis?



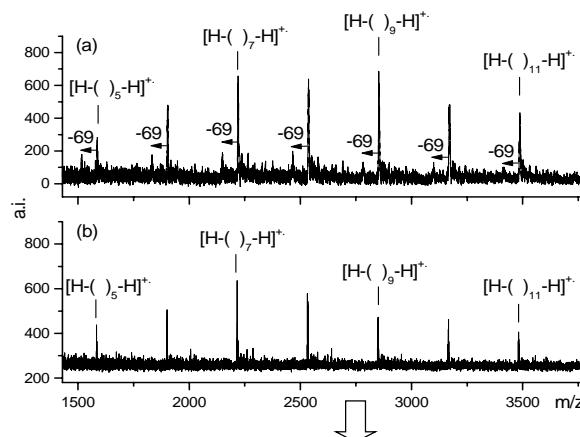
oligomerization degree $\sim n = 25$



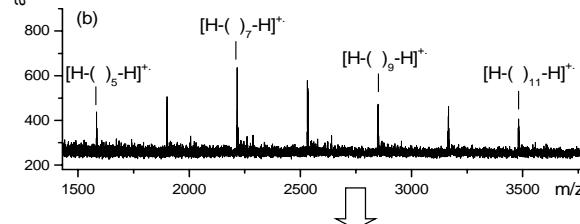
Influence of the MALDI Matrix



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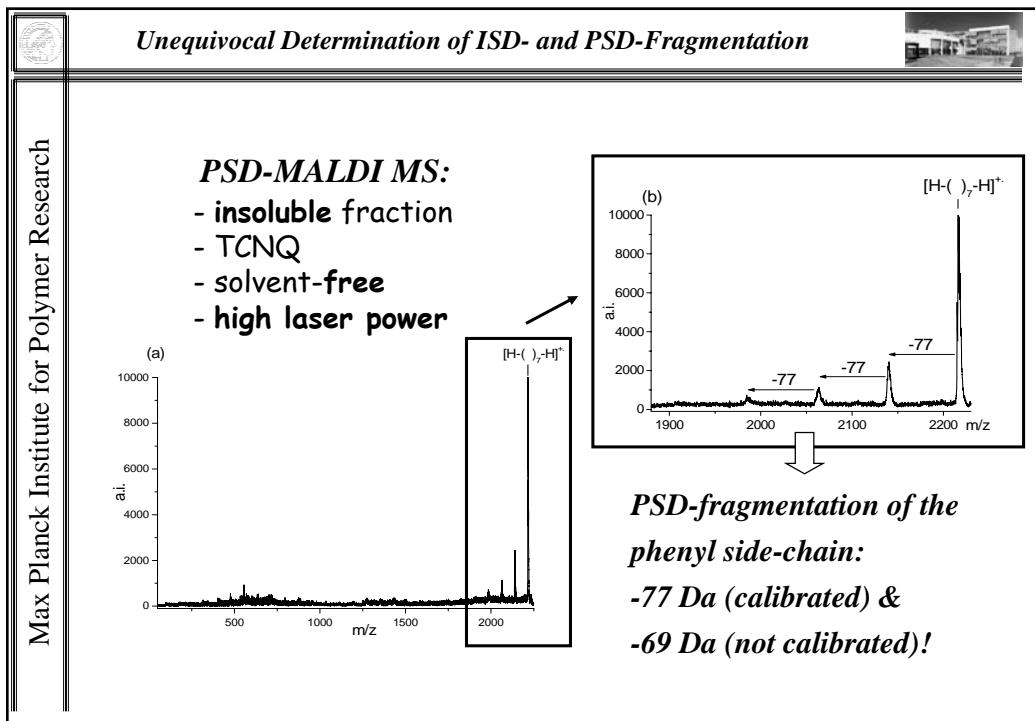
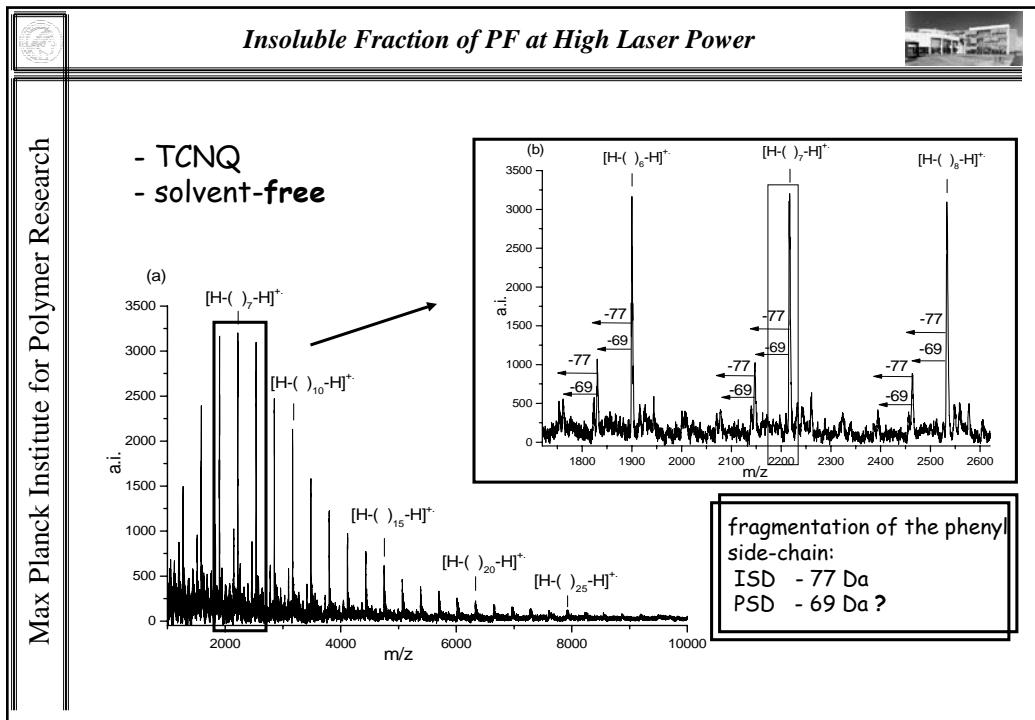


- insoluble fraction
- solvent-free
- minimum
- laser power
- dithranol



- insoluble fraction
- solvent-free
- minimum
- laser power
- TCNQ

*successful suppression of fragmentation
& independence of matrix choice!*





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Traditional Copolymer Characterization:

- ❖ combinations of analytical methods (reactivity ratios, NMR)
- ❖ averaged properties of different chain lengths

MALDI-MS Copolymer Characterization:

- ❖ Allows *direct* information of
- Copolymer distribution and composition

Information about the primary structure?

- ❖ „random“- and „block“-polymer sequenz determination



MALDI-*Fragmentation MS?*



Composition of a diblock copolymer

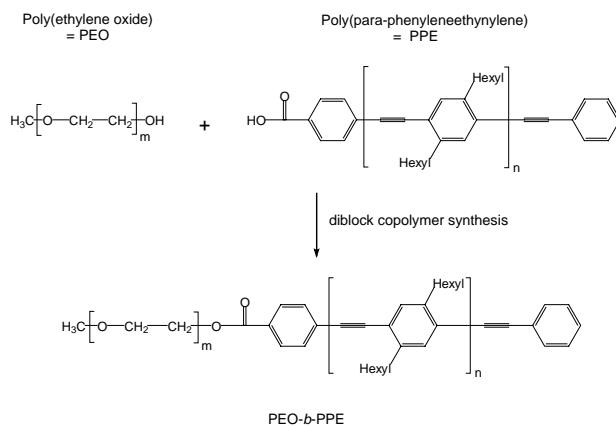


Compositional information of resolved molecular weight distributions are not unequivocal

Interpretation of mass spectra often unclear
due to signal overlapping caused by :

- unfavorable monomer masses e.g. $M_{\text{Monomer1}} \sim k \times M_{\text{Monomer2}}$
- low resolution due to high mass range or instrumental limitation

Formation of a PPE-*b*-PEO diblock copolymer



$$\begin{aligned} M_{\text{copo}} &= M_{\text{End1}} + m \times M_{\text{Monomer1}} + M_{\text{Spacer}} + n \times M_{\text{Monomer2}} + M_{\text{End2}} \\ &= 15 + m \times 44 + 120 + n \times 268 + 101 \end{aligned}$$

*Composition of a PPE-*b*-PEO diblock copolymer*

The figure illustrates the fragmentation of a PPE-*b*-PEO diblock copolymer. The main plot shows the MALDI-TOF mass spectrum with m/z from 1400 to 3400. Key peaks are labeled: $M_{Monomer1} = 44$, $268 = M_{Monomer2}$, and a base peak at $m/z = 44$. A bracket indicates the range from $m/z = 1900$ to 2100 . Two zoomed-in plots are shown: one for the m/z range 1900-2200 labeled "pure signal" which highlights peaks at $2007 = M_{copo}$ and $2007 - \Delta M = 5$; and another for the m/z range 2100-2400 labeled "overlapping signals" which highlights peaks at $2227 = M_{copo}$, $2227 - \Delta M = 8$, $2231 = M_{copo}$, and $2231 - \Delta M = 14$.

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$M_{copo} = M_{End1} + m \times M_{Monomer1} + M_{Spacer} + n \times M_{Monomer2} + M_{End2}$

$$= 15 + m \times 44 + 120 + n \times 268 + 101$$

i.

$M_{Monomer1} = 44$

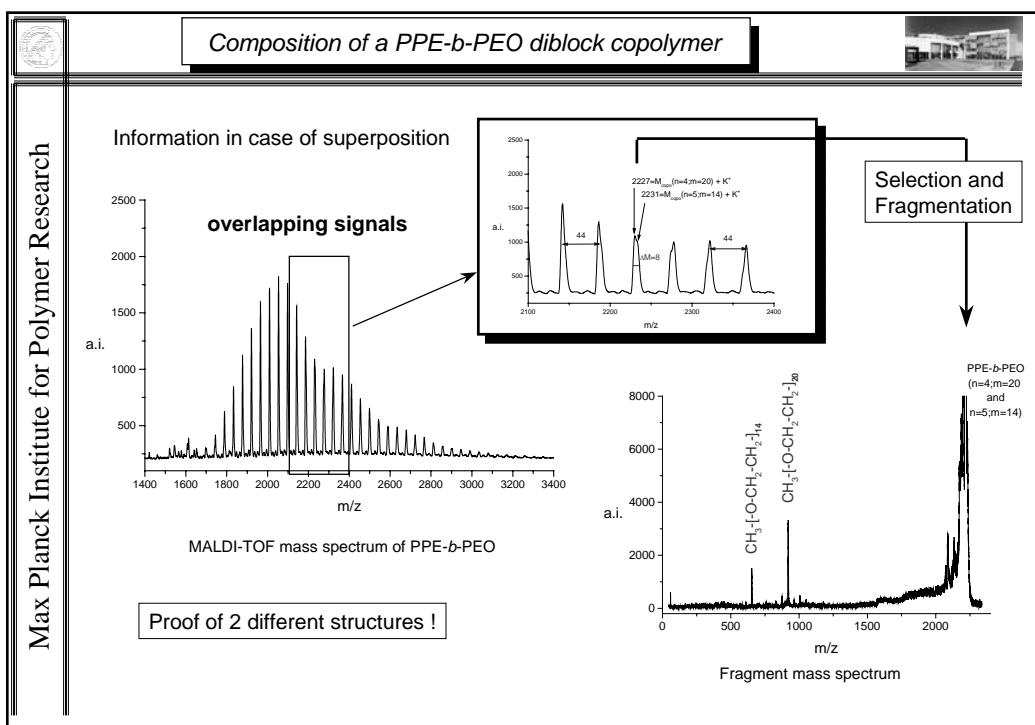
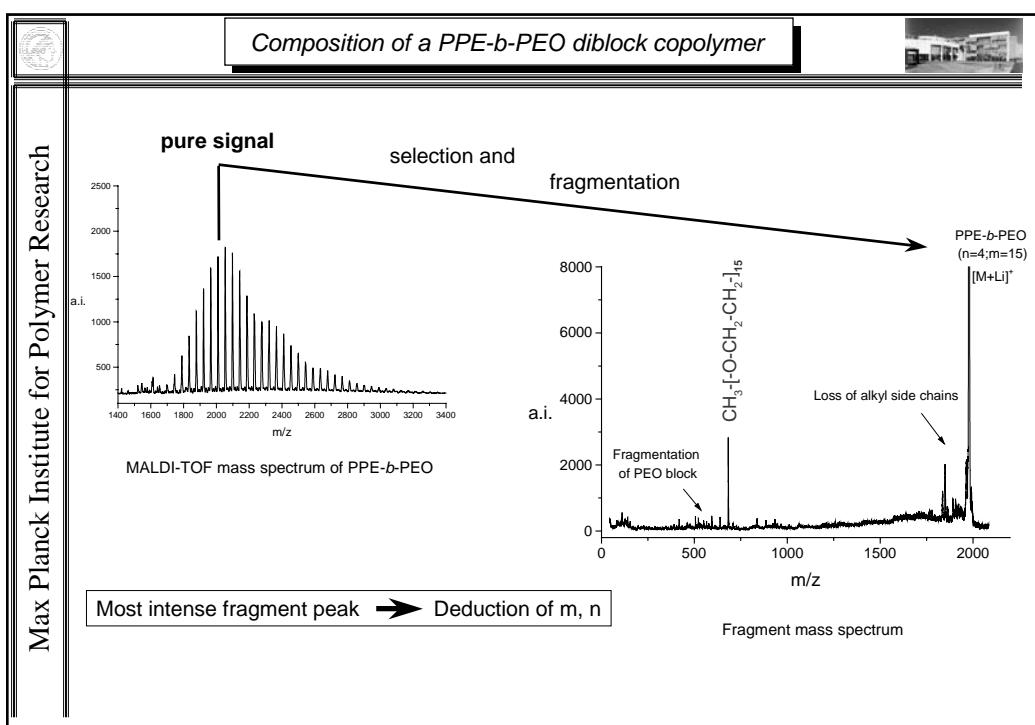
$268 = M_{Monomer2}$

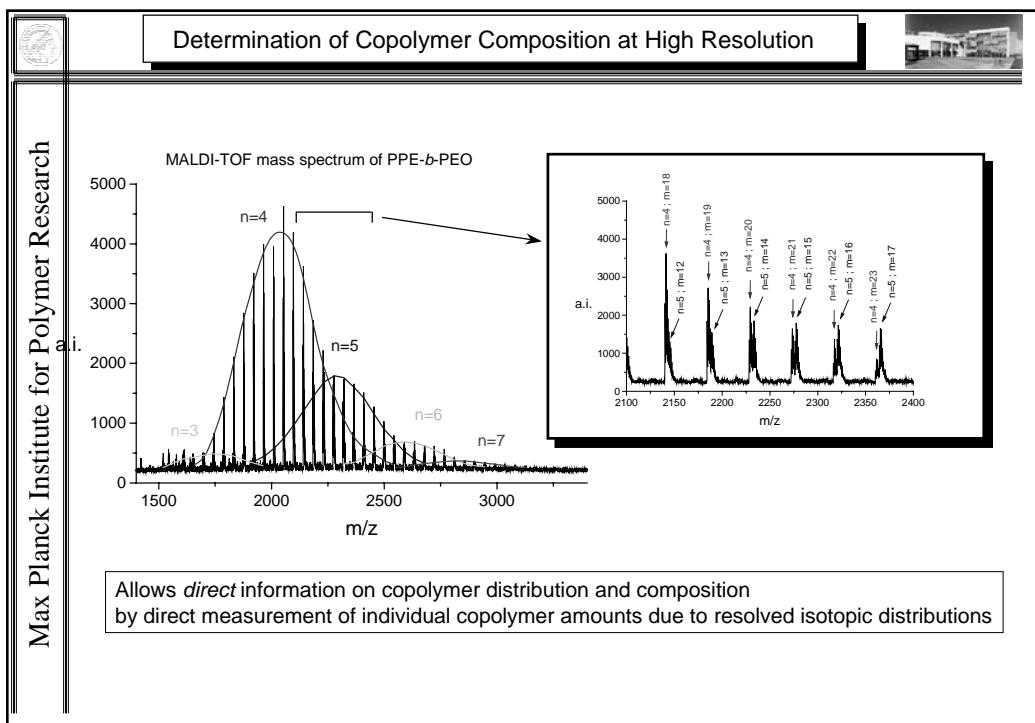
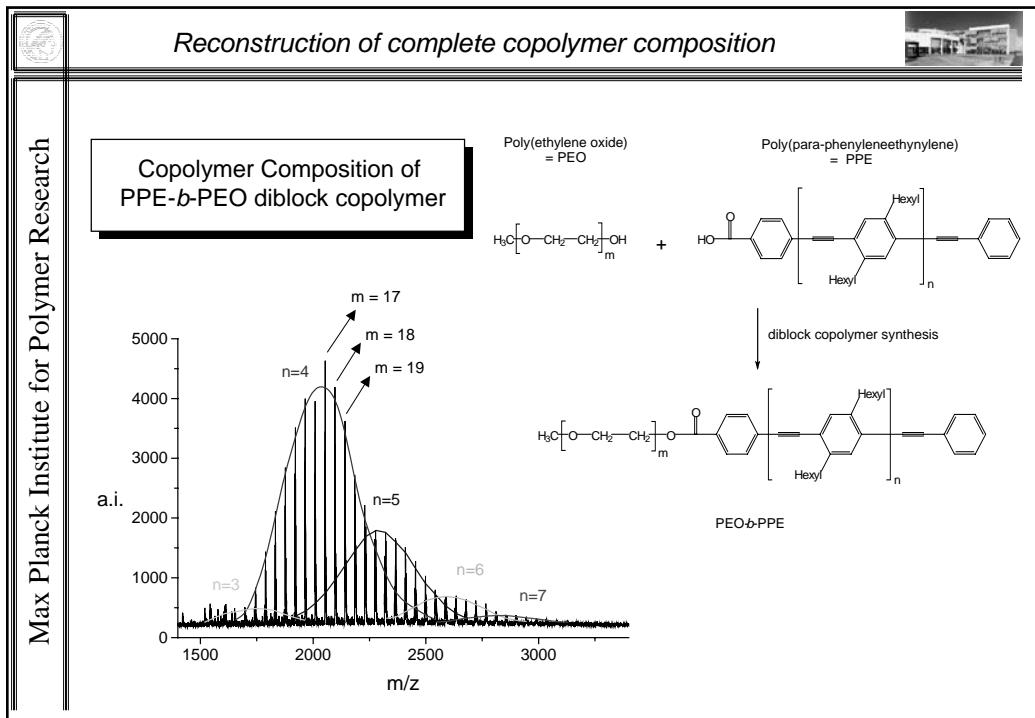
MALDI-TOF mass spectrum of PPE-*b*-PEO (low resolution)

pure signal

overlapping signals

Need of additional data → fragmentation of copolymer







Sequence Determination of Synthetic Polymers

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“Copolymer”:

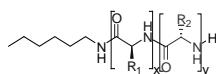
❖ Monomer Units: 2

❖ Copolymer Types:

❖ A B A B B A “random”

❖ A A A B B B “block”

❖ amino acid monomer units



❖ no given primary structure

❖ polydisperse

❖ protected side-chains

Characteristical structural elements of
natural polypeptides and industrial polymers



Synthetic Polypeptides

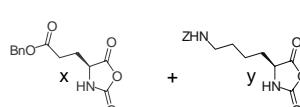
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„Random“ Copolymers:

$n\text{-Hex-NH-(Glu)}_x\text{-st-(Lys)}_y\text{-H}$

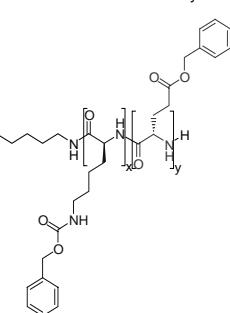
Glu

Lys



$n\text{-C}_6\text{H}_{13}\text{NH}_2$
DMF, RT
-CO₂

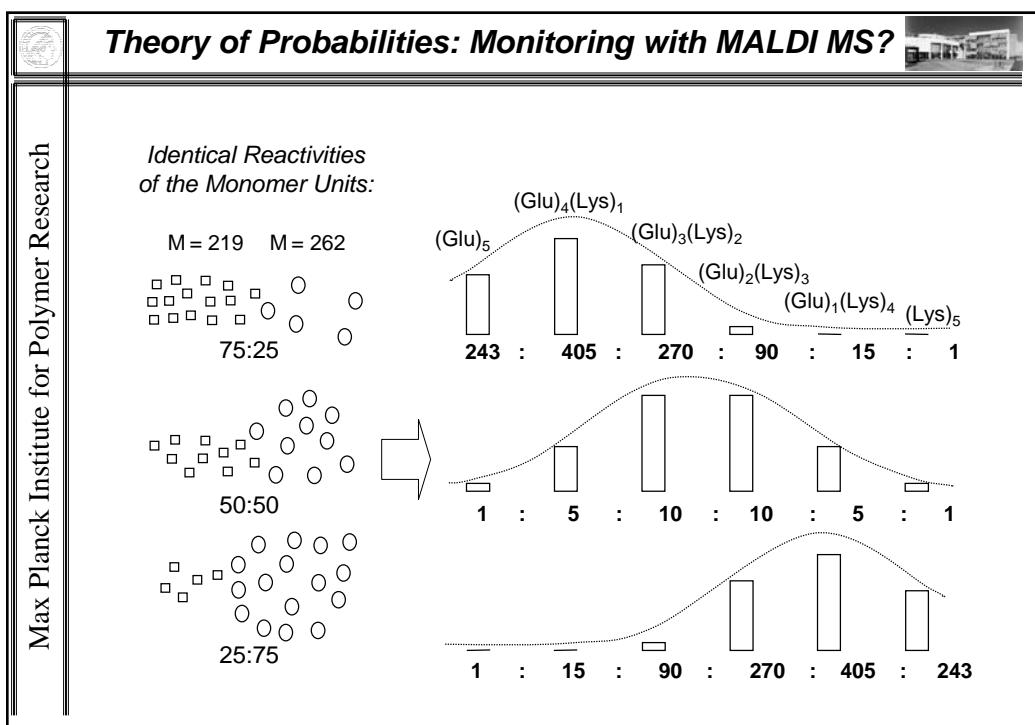
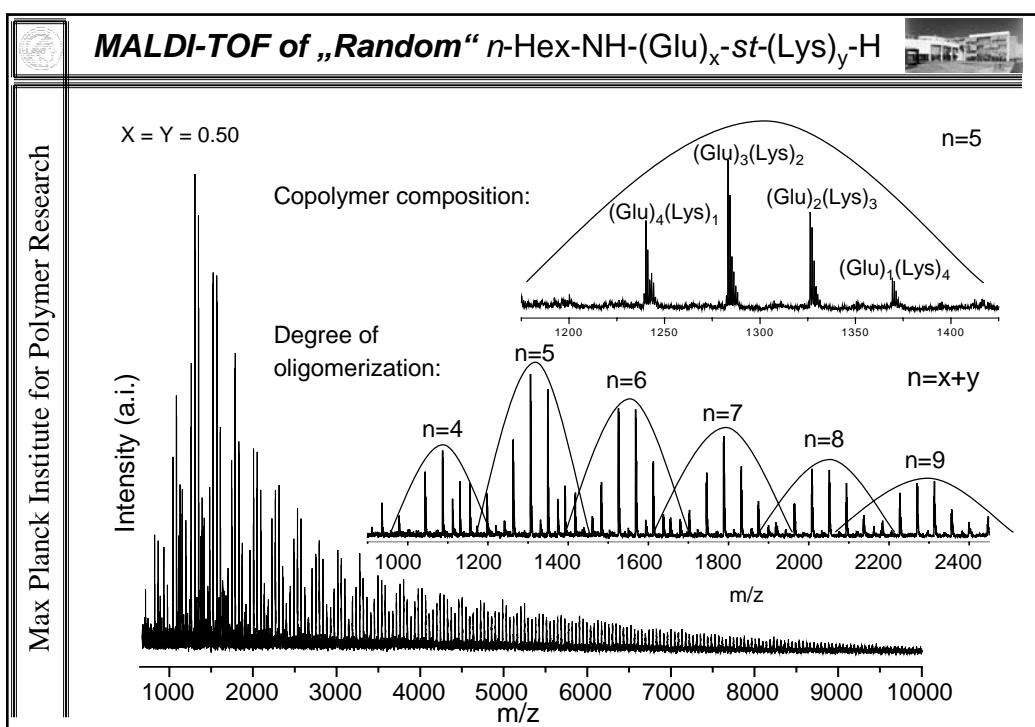
Reactivity: Glu ≈ Lys

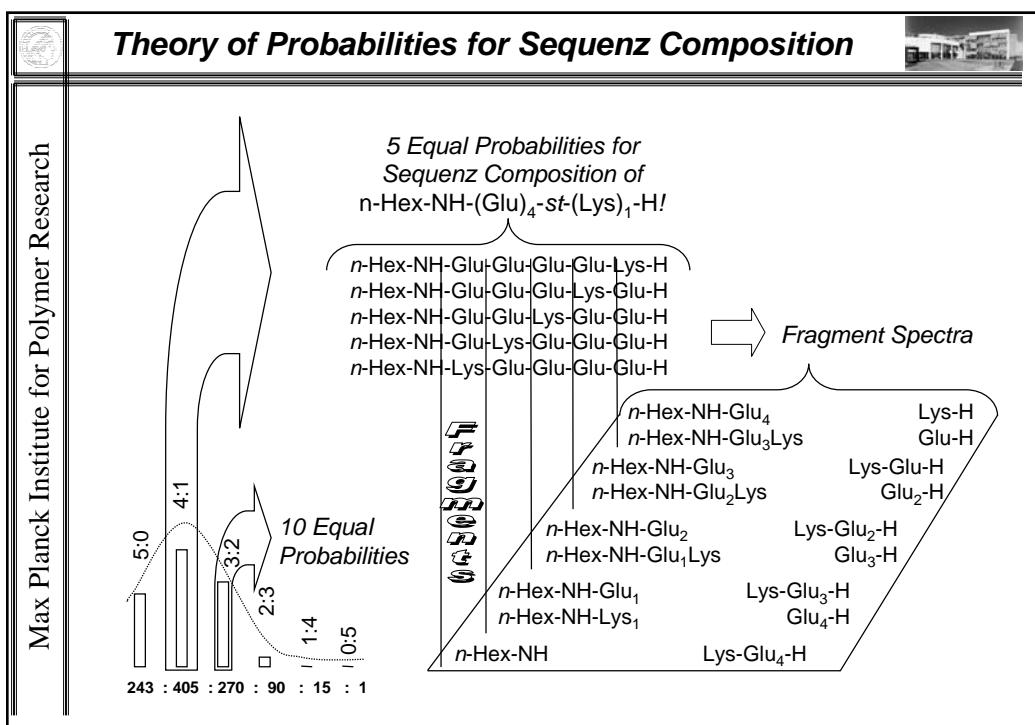
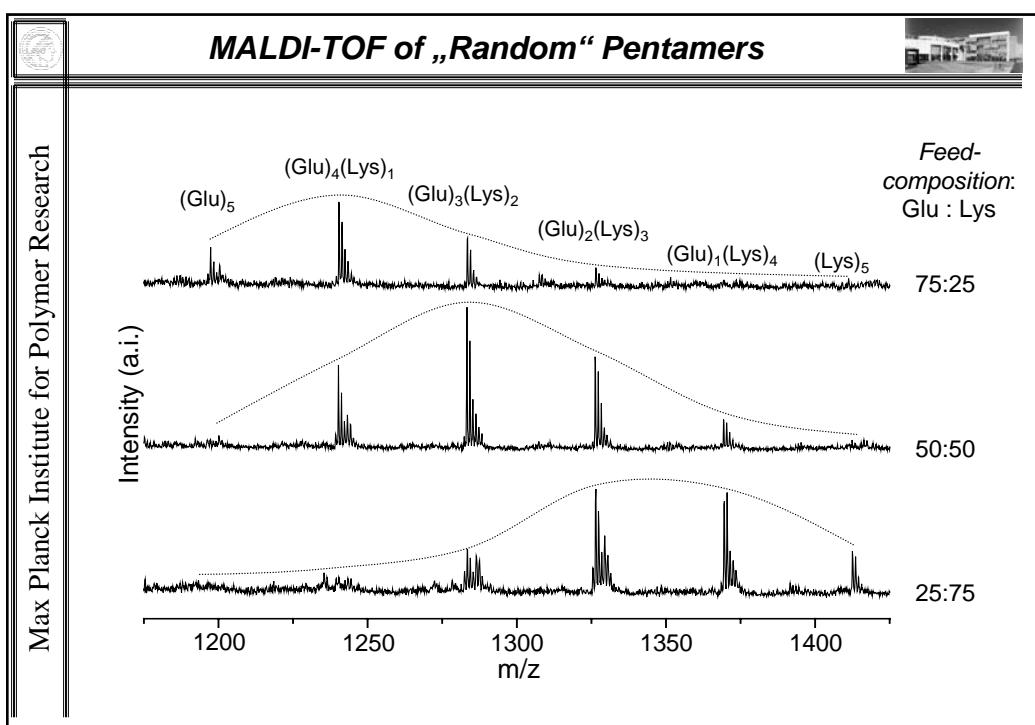


Block-Copolymers:

$n\text{-Hex-NH-(Glu)}_x\text{-bl-(Lys)}_y\text{-H}$

$n\text{-Hex-NH-(Lys)}_y\text{-bl-(Glu)}_x\text{-H}$



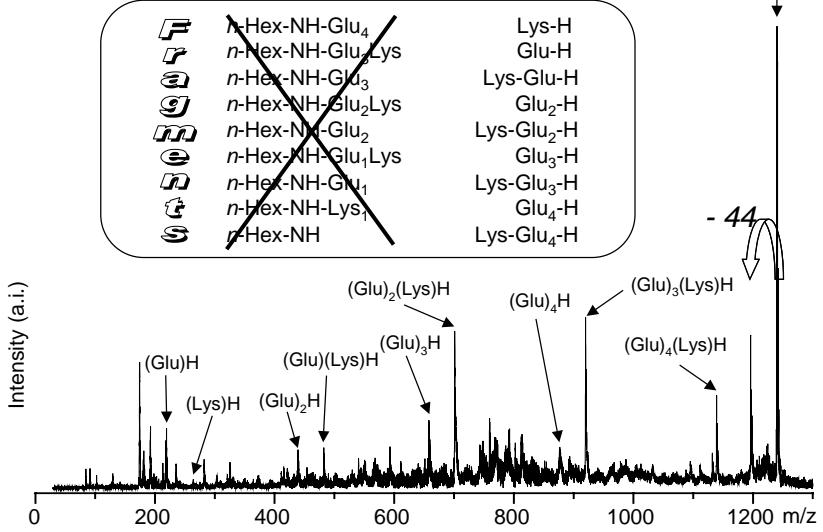
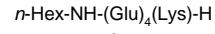




PSD Fragment Ion-Analysis of a „Random“ Pentamer



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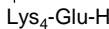
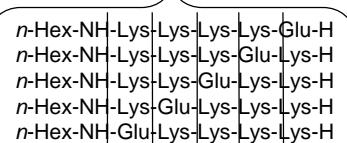


Sequenz analysis of a Block-Copeptid

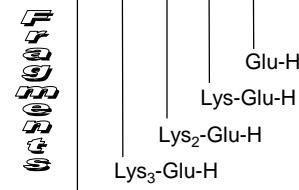
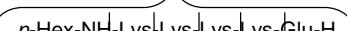


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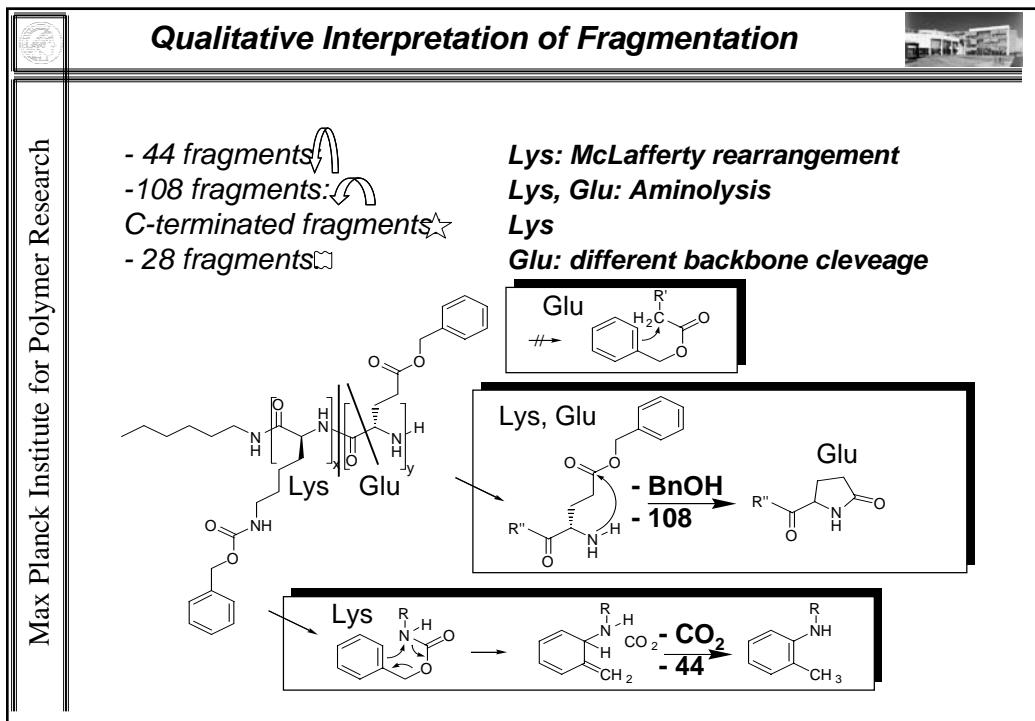
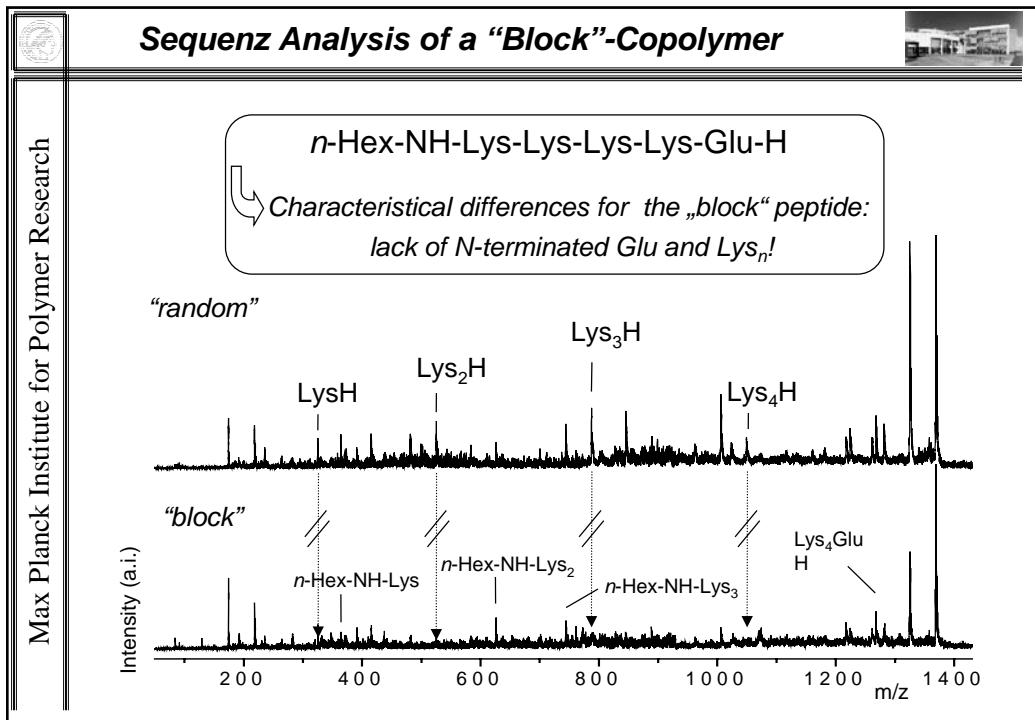
Theory of Probabilities:



Pre-determined Sequenz:



Characteristical differences!





Acknowledgement



Coworkers

Laurence Przybilla ¹

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Sabine Kummer

Kai Martin

Kimihiro Yoshimura

Jochen Spickermann

Ali Pouhanipour

Stefan Türk